





Journal of the European Ceramic Society 27 (2007) 1059–1063

www.elsevier.com/locate/jeurceramsoc

MgWO₄, ZnWO₄, NiWO₄ and CoWO₄ microwave dielectric ceramics

R.C. Pullar*, S. Farrah, N. McN. Alford

Centre for Physical Electronics and Materials, Faculty of Engineering, Science and the Built Environment, London South Bank University, 103 Borough Road, London SEI 0AA, UK

Available online 19 June 2006

Abstract

AWO₄ (where A = Zn²⁺, Mg²⁺, Ni²⁺ and Co²⁺) tungstate ceramics were synthesised from a standard ceramic route, and their sintering behaviour and microwave dielectric properties were investigated. All AWO₄ powders formed single-phase materials, except MgWO₄, and they all had a wolframite-like monoclinic P2/c (13) structure, not the tetragonal scheelite structure found in CaWO₄. MgWO₄ had a poor Qf which deteriorated with further sintering, due to discontinuous grain growth at relatively low temperatures of $1000\,^{\circ}$ C and over. At higher temperatures, ZnWO₄, NiWO₄ and CoWO₄ have high Qf values at $1200\,^{\circ}$ C of 62,800,24,900 and 38,600 GHz, respectively, which will all almost certainly be improved with firing at higher temperatures, particularly in the case of CoWO₄. Although not at all well sintered, ZnWO₄ and CoWO₄ have high Qf values of 34,000 and 28,900 GHz, respectively, at temperatures below the LTCC limit. © 2006 Elsevier Ltd. All rights reserved.

Keywords: Sintering; Powders-solid state reaction; Microstructure-final; Dielectric properties; Tungstates

1. Introduction

With the continuing growth of mobile telecommunications, there is always an interest in novel ceramics for use as dielectric resonators (DRs) at microwave frequencies (1-20 GHz). An effective DR should have a sufficiently high relative permittivity to allow miniaturisation of the component ($\varepsilon_r > 10$) and low dielectric losses at microwave frequencies to improve selectivity (Q > 5000, where $Q = 1/\tan \delta$). Microwave ceramics are currently available with low τ_f and Qf > 100,000 ($Qf = Q \times f_r$). However, these are usually made from complex perovskites, such as the mixed metal tantalate perovskites BaZn_{0.33}Ta_{0.67}O₃ (BZT) and BaMg_{0.33}Ta_{0.67}O₃ (BMT).²⁻⁴ All complex perovskites require high sintering temperatures (>1400 °C), and the structures and properties of the complex perovskites (often with four or more cations) are proving difficult to predict, and depend strongly upon the degree of ordering.⁵ The next generation of dielectric ceramic components need to be miniaturised, multilayer devices, packaged in conjunction with conducting layers, with minimal dielectric losses. This has lead to a recent growth in the study of dielectric low temperature co-fired ceramic (LTCC) materials. LTCC technology provides a promising way to realise

multi-chip modules and 3D circuits, for use in mobile telecommunication devices and broadband systems. However, to achieve this a new generation of DR materials must be developed, that sinter at sufficiently low temperatures (<960 °C, the melting point of Ag, or <1085 °C if using copper), still have the required dielectric properties at microwave frequencies, and are inexpensive.

The AWO₄ compounds are well known. The most common are the minerals scheelite (CaWO₄), used for its optical properties (scintillation counters, lasers, optical fibres),⁶ and wolframite ([Fe/Mn]WO₄), the most common raw material for producing tungsten metal.⁷ In the A²⁺WO₄ compounds, if A²⁺ has a small ionic radius <0.77 Å (Mg, Zn) it will have the monoclinic wolframite structure, but larger A²⁺ cations (Ca, Ba) form the tetragonal scheelite structure. 8 Monoclinic ZnWO₄ is also a known mineral, named sanmartinite. Recently, Kim et al. reported on the microwave properties of CaWO₄ at the Third International Conference on Microwave Materials and their Applications (MMA 2004) conference in October 2004, producing CaWO₄ with a low ε_r of 9.7 but a high Qf of 44,200 GHz.⁹ Furthermore, Kim et al. reported that when doped with 0.5 wt.% Bi₂O₃-9 wt.% HBO₃ CaWO₄ could be sintered at only 850 °C with an improved Qf of 70,000 GHz. 10 Also, at the same conference, Choi et al. reported that CaMoO₄ (also scheelite structure) had $Qf = 55,000 \,\text{GHz}$ and $\varepsilon_r = 11.7$, when sintered at $1100 \,^{\circ}\text{C}$ with a slight 1.02 excess of Mo.¹¹

^{*} Corresponding author. Tel.: +44 20 7815 7540; fax: +44 20 7815 7699. E-mail address: r.c.pullar@lsbu.ac.uk (R.C. Pullar).

However, no one has looked at the microwave dielectric properties of any other related AWO₄ or AMoO₄ compounds. To this end series of ceramics based upon wolframite (CaWO₄) and scheelite (CaMoO₄) were made and characterised by microwave measurements, dilatometry, XRD and SEM. This paper details the synthesis and investigation of A²⁺WO₄ ceramics where A²⁺ = Mg, Zn, Co and Ni, their sintering behaviour, and quality factor (Q), and dielectric permittivity ($\varepsilon_{\rm r}$) at microwave frequencies. It was also hoped that some of these materials could be useful candidates for LTCC dielectrics. Results on A²⁺MoO₄ compounds will be reported in the future.

2. Experimental

All A²⁺WO₄ samples were prepared by a standard ceramics mixed-oxide route (oxides atleast 99% pure), from MgO, WO₃, ZnO, NiO and Co₃O₄. A stoichiometric mix of the oxides needed to form each compound was ball milled in deionised water with zirconia milling media for 1 day to mix the powders, and then dried on a rotary evaporator. The resultant powder was then calcined at 900 °C for 12 h in air, and then milled in water at 300 rpm/3 h on Fritsch Pulverisette high-energy planetary ball mill to reduce particle size to $<3 \mu m$. The resultant fine powders were again dried on a rotary evaporator, and then used to make all samples, without any further processing or addition of an organic binder. The powders were uniaxially pressed to form pellets under a pressure of 100 MPa using a 13 mm diameter die. All samples were then sintered in air, at temperatures between 900 and 1200 °C/2 h, heated and cooled at a rate of 5 °C min⁻¹. The density of the samples was calculated from their measured volume and mass, and theoretical maximum densities were obtained from the Philips JCPDF files.

The change in length of a cylindrical sample, dL/L_0 , with increasing temperature was measured at a rate of $5\,^{\circ}\mathrm{C}\,\mathrm{min}^{-1}$ using a Netzsch 402C dilatometer. This was expressed as a % change in length with temperature, to study the sintering behaviour of the samples.

X-ray diffraction patterns of the samples treated at various temperatures were recorded in the region of $2\theta = 20-75^{\circ}$ on a PANalytical X'Pert MRD diffractometer using Cu K_{α} radiation, with a PANalytical X'cellerator detector. The results were analysed and manipulated using PANalytical X'Pert software.

The microstructure and grain size of the samples were investigated using a field emission electron microscope (Hitachi S-4300), on unpolished samples.

The quality factor (Q) and relative permittivity (ε_r) were measured at frequencies of 8.5–10 GHz, using the resonant TE_{0.1.8} mode of the sample, ¹² in an oxygen-free high-conductivity copper cavity, whose diameter was three times that of the sample, as recommended by Kajfez and Guillon. ¹² The TE_{0.1.8} mode was examined using a Hewlett-Packard HP8720D Vector Network Analyser, and the surface resistance of the copper was calculated from the Q value of the TE_{0.1.1} resonance of the empty cavity, to allow the results to be corrected for any loss due to the cavity walls. ¹² The Q values are corrected for losses due to the measurement equipment, and so can be assumed to be the Q of the dielectric ceramic. Measurements were made on the as-fired

samples, at room temperature. The results are given as Qf, that is $Q \times \text{resonant}$ frequency (f_r) , and f_r was between 8.5 and 10 GHz in all cases.

3. Results and discussion

There are few reports on the synthesis of these materials, but NiWO₄ and ZnWO₄ have been made at 700 °C from amorphous powders, 6 and W⁶⁺ has been shown to diffuse well in CaWO₄ at 900 °C, 13 so it was decided to calcine the oxides at 900 °C for 12 h. The XRD patterns of the calcined powders are shown in Fig. 1. As expected, all formed a monoclinic wolframite-like structure, with space group P2/c (13). All were pure single phase, except for MgWO₄, which had a small amount of an unidentified second phase with a peak at 37.5°. The ZnWO₄ and MgWO₄ powders were white, while NiWO₄ was deep yellow and CoWO₄ was a rich blue colour.

The shrinkage behaviour of the pressed powders is shown in Fig. 2, heated continuously at 5 °C min⁻¹. None have sintered below the LTCC target temperatures of 1050 °C (Cu) or 950 °C (Ag), and MgWO₄ does not sinter much at all below 1250 °C. At that temperature the material suddenly expands and de-densifies, typical of a phase change, and any MgWO₄ pellets fired to this temperature disintegrate into a fine powder. Both ZnWO₄ and NiWO₄ appear fairly well sintered by 1200 °C, whereas CoWO₄ has barely begun sintering at this temperature, but then rapidly sinters over the next hundred degrees, seemingly needing temperatures of over 1250 °C to become dense. Samples were fired between 900 and 1200 °C, and held for 2 h at that temperature, and the changes in density are shown in Fig. 3. It can be seen that the density of MgWO₄ levels out and remains roughly constant above 1050 °C, at a density of around 6.00 g cm⁻³, only 87.0% of the maximum theoretical density (MTD = 6.896 g cm^{-3}). ZnWO₄ (MTD = 7.872 g cm^{-3}) and $NiWO_4$ (MTD=7.954 g cm⁻³) have low densities of $4.47 \,\mathrm{g \, cm^{-3}}$ (56.8%) and $4.90 \,\mathrm{g \, cm^{-3}}$ (61.6%), respectively, at 900 °C, but they sinter steadily up to $6.86 \,\mathrm{g \, cm^{-3}}$ (87.1%) and 7.29 g cm⁻³ (91.7%) at 1200 °C, and should sinter further at higher temperatures. $CoWO_4$ (MTD=7.760 g cm⁻³) has only increased in density from 4.40 g cm⁻³ (56.7%) at 1050 °C, the lowest firing temperature at which the pellets were sufficiently reacted to be handleable, to $5.39 \,\mathrm{g}\,\mathrm{cm}^{-3}$ (69.5%) at $1200 \,^{\circ}\mathrm{C}$. The dilatometer results suggest that CoWO4 requires atleast a further 50 °C rise in firing temperature to become a well sintered sample. Unfortunately, due to extreme furnace contamination issues from vaporisation of WO₃, MoO₃ and other components from AWO₄ and AMoO₄ samples above 1200 °C, we have had to temporarily suspend firing these samples above 1200 °C.

The plots of ε_r and Qf against sintering temperature are shown in Figs. 4 and 5. As expected, ε_r closely follows density in all cases, and is higher than CaWO₄ ($\varepsilon_r = 9.7$)⁹ for all well sintered samples, reaching 12.1 for MgWO₄, 13.5 for ZnWO₄ and 13.3 for NiWO₄. Even the poorly sintered CoWO₄ achieves ε_r higher than CaWO₄, with a value of 10.7, in a ceramic that is only 69.5% dense.

For all MgWO₄ samples Qf is low, but unusually with a maximum of 5400 GHz in the least sintered sample at 950 $^{\circ}$ C (only

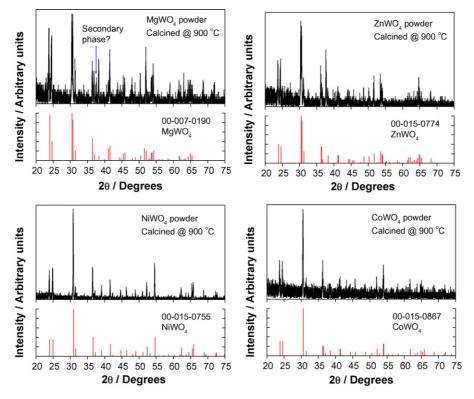


Fig. 1. XRD patterns of calcined Mg, Zn, Ni and CoWO₄ powders.

71.2% dense). After this poor start, the Qf then decreases with increasing temperature and density to under 1000 GHz. This can be explained by the microstructure—the sample appears well sintered at 950 °C, despite its low density (Fig. 6(a)), but with further increases in temperature large grain growth occurs, resulting in the existence of some very large grains due to discontinuous grain growth (DGG) (Fig. 6(b)). This DGG has been shown to be the cause of very poor Q in over sintered niobate

ceramics, ¹⁴ and once it occurs Q values can decline sharply even as density and ε_{Γ} continue to rise.

ZnWO₄ has good Qf values even at low temperatures (26,000 GHz at 950 °C), and very high Qf at 1200 °C (63,000 GHz). Almost no grain growth has occurred between 950 and 1200 °C, and density is <90%, so Q may still improve with further sintering. NiWO₄ has low Qf values at lower temperatures, achieving a poor minimum of only 3200 GHz at 1050 °C, but Qf rises suddenly above 1150 °C to 25,000 GHz. Again, this can be explained by the microstructure—the ceramic is clearly

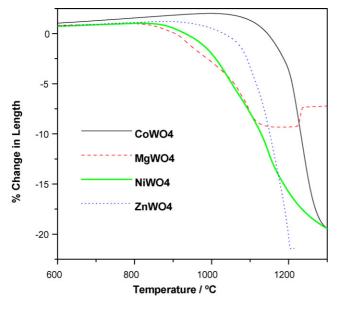


Fig. 2. Shrinkage measurements of AWO₄ pellets pressed from calcined powders, fired at $5\,^{\circ}\text{C}$ min $^{-1}$.

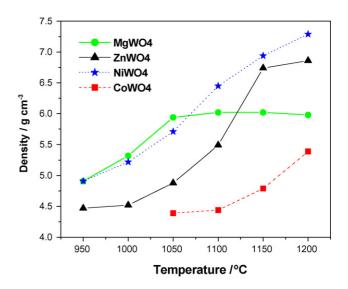


Fig. 3. Variation of density with sintering temperature for AWO_4 ceramics.

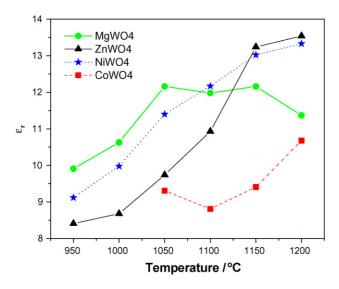


Fig. 4. Variation in ε_r with firing temperature for AWO₄ ceramics.

poorly sintered at 950 °C, and remains so at 1150 °C, despite achieving 88% density (Fig. 7(a) and (b)). The ceramic is better sintered at 1200 °C (Fig. 7(c)), but shows little grain growth, indicating that higher Qf values may well be attainable at higher

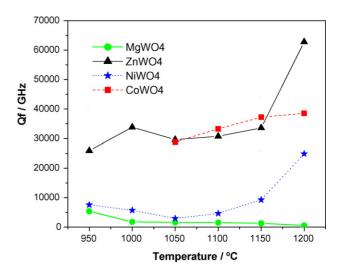


Fig. 5. Variation in Qf with firing temperature for AWO₄ ceramics.

sintering temperatures. CoWO₄ has a surprisingly high *Qf* for such a poorly sintered ceramic—up to 39,000 GHz in a sample only 69.5% dense at 1200 °C, which can certainly be improved upon with higher firing temperatures.

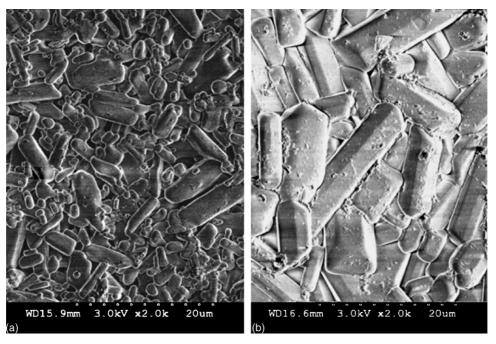


Fig. 6. SEM micrographs (2K magnification) of unpolished MgWO₄ fired to (a) $950\,^{\circ}\text{C/2}\,\text{h}$ and (b) $1200\,^{\circ}\text{C/2}\,\text{h}$.

Table 1
Properties of AWO₄ samples with the highest *Qf* values

| Material | Temperature (°C) (fired 2 h) | Qf (GHz) | ε_{r} | Density (g cm ⁻³) | X-ray density ^a | Maximum density (%) |
|--------------------------------|------------------------------|----------|----------------------------|-------------------------------|----------------------------|---------------------|
| MgWO ₄ | 950 | 5400 | 9.9 | 4.91 | 6.896 | 71.2 |
| $ZnWO_4$ | 1200 | 62800 | 13.5 | 6.86 | 7.872 | 87.1 |
| $NiWO_4$ | 1200 | 24900 | 13.3 | 7.29 | 7.954 | 91.7 |
| $CoWO_4$ | 1200 | 38600 | 10.7 | 5.39 | 7.760 | 69.5 |
| CaWO ₄ ^b | 1200 | 44200 | 9.7 | n/a | 6.120 | n/a |

^a From Philips JCPDF files.

^b From Kim et al.,⁹ included here for comparison.

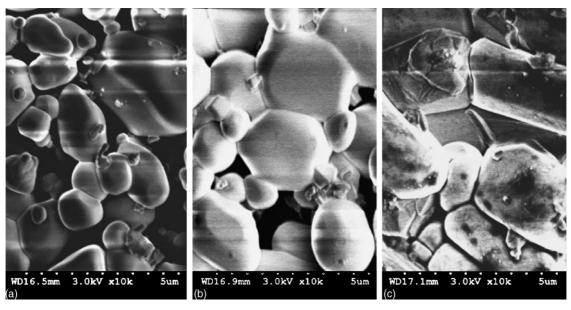


Fig. 7. SEM micrographs (10K magnification) of unpolished NiWO₄ fired to (a) $950\,^{\circ}$ C/2 h, (b) $1150\,^{\circ}$ C/2 h and (c) $1200\,^{\circ}$ C/2 h.

It should be noted that although they are not at all well sintered, $ZnWO_4$ and $CoWO_4$ have high Qf values of 34,000 and 28,900 GHz, respectively, at temperatures below the LTCC limit, although porosity could be a problems regarding any possible applications of such materials. Table 1 shows the properties of the sample of each ceramic with the highest Qf values, compared to $CaWO_4$ for comparison.

4. Conclusion

All AWO₄ powders were formed as single-phase materials, except maybe MgWO₄ which had a small amount of an unidentified second phase, and they all had a wolframite-like monoclinic P2/c (13) structure. All sintered above the LTCC temperature limits, but although not at all well sintered, ZnWO₄ and CoWO₄ have high *Qf* values of 34,000 and 28,900 GHz, respectively, at temperatures below the LTCC limit. MgWO₄ had a poor *Qf* which deteriorated with further sintering, due to discontinuous grain growth at relatively low temperatures of 1000 °C and over. At 1250 °C MgWO₄ undergoes a phase change, resulting in disintegration of the sintered ceramic sample to a fine powder. At higher temperatures, ZnWO₄, NiWO₄ and CoWO₄ have high *Qf* values developing at 1200 °C of 62,800, 24,900 and 38,600 GHz, respectively, which will all almost certainly be improved with firing at higher temperatures, particularly in the case of CoWO₄.

References

- Wersing, W., Microwave ceramics for resonators and filters. Curr. Opin. Solid State Mater. Sci., 1991, 1, 715–731.
- Tamura, H., Konoike, T., Sakabe, Y. and Wakino, K., Improved high-Q dielectric resonator with complex structure. *J. Am. Ceram. Soc.*, 1984, 67, C59–C61.

- Kawashima, S., Nishida, N., Ueda, I. and Ouchi, H., Ba(Zn_{1/3}Ta_{2/3})O₃ ceramics with low dielectric loss at microwave-frequencies. *J. Am. Ceram. Soc.*, 1983, 66, 241–243.
- Matsumoto, K., Hiuga, T., Takada, K. and Ichimura, H., Ba(M_{1/3}Ta_{2/3})O₃ ceramics with ultra-low loss at microwave-frequencies. *IEEE Trans. Ultra-son. Ferroelectrics Freq. Control*, 1986, 33, 802.
- Akbas, M. A. and Davies, P. K., Ordering-induced microstructures and microwave dielectric properties of the Ba(Mg_{1/3}Nb_{2/3})O₃–BaZrO₃ system. *J. Am. Ceram. Soc.*, 1998, 81, 670–676.
- Kuzmin, A. and Purans, J., Local atomic and electronic structure of tungsten ions in AWO₄ crystals of scheelite and wolframite types. *Rad. Meas.*, 2001, 33, 583–586.
- 7. http://mineral.galleries.com/.
- Sleight, A. W., Accurate cell dimensions for ABO₄ molybdates and tungstates. Acta Crystallogr. B, 1972, 28, 2899–2902.
- Kim, J. S., Lee, J. C., Cheon, C. I. and Lee, C. H., New type of wolframite LiYW₂O₈ for LTCC applications. In *Proceedings of the Third International* Conference on Microwave Materials and their Applications (MMA 2004), 2004.
- Kim, E.-S., Kim, S.-H. and Lee, B. I., Low temperature sintering of CaWO₄ ceramics for LTCC applications. In *Proceedings of the Third International Conference on Microwave Materials and their Applications (MMA 2004)*, 2004
- Choi, G.-K., Cho, A.-Y., An, J.-S. and Hong, K. S., Microwave dielectric properties and sintering behaviours of scheelite compound CaMoO₄. In Proceedings of the Third International Conference on Microwave Materials and their Applications (MMA 2004), 2004.
- Kajfez, D. and Guillon, P., *Dielectric Resonators*. Artech House, Zurich, Switzerland, 1986, pp. 327–376.
- Guo, Q. and Kleppa, O. J., Enthalpies of formation from the component oxides of MgWO₄, CaWO₄, SrWO₄ and BaWO₄, determined by high temperature direct synthesis calorimetry. *Thermochim. Acta*, 1996, 288, 53–61.
- Pullar, R. C., Breeze, J. D. and Alford, N. McN., Microwave dielectric properties of columbite-structure niobate ceramics, M²⁺Nb₂O₆. Key Eng. Mater., 2002, 224–226, 1–4.