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Short communication

A novel low firing microwave dielectric ceramic NaCa₂Mg₂V₃O₁₂

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Abstract

A novel low-firing microwave dielectric ceramic NaCa₂Mg₂V₃O₁₂ was prepared by the conventional solid state ceramic route. The phase composition, microstructure, and sintering behavior were investigated. The NaCa₂Mg₂V₃O₁₂ ceramic could be well densified at around 915 °C and exhibited favorable microwave dielectric properties with a low relative permittivity of ~10, a high quality factor (Q $Q_u \times f$) of ~50,600 GHz, and a negative temperature coefficient of resonant frequency (τ_f) of ~47 ppm/°C. The NaCa₂Mg₂V₃O₁₂ ceramic can be compatible with Ag electrode, which makes it a promising ceramic for LTCC technology application. CaTiO₃ was added into the NaCa₂Mg₂V₃O₁₂ ceramic to obtain a near zero τ_f , and 0.8NaCa₂Mg₂V₃O₁₂–0.2CaTiO₃ ceramic sintered at 930 °C for 4 h showed improved properties with ε_r =11.5, $Q_u \times f$ =37,500 GHz, and τ_f =2 ppm/°C.

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1. Introduction

Low-temperature co-fired ceramics (LTCC) has been generating considerable interest due to the benefits offered for the fabrication of miniature multilayer devices [1–4]. A typical LTCC module is a multilayered compact of ceramic substrates and common internal electrode material Ag which will melt at 961 °C. Advanced substrate materials for microwave integrated circuits should possess a high quality factor ($Q_u \times f$), a small temperature coefficient of resonant frequency (τ_f), and especially a low relative permittivity ($\varepsilon_r < 15$) to avoid signal delay [5].

Several ceramics such as Al_2O_3 [6], Mg_2SiO_4 [7], $Mg_4Nb_2O_9$ [8], and $MgSiO_3$ [9] have been developed and regarded as low ε_r microwave dielectric ceramics. However, these ceramics suffer the disadvantage of high sintering temperatures. In order to reduce the sintering temperature of dielectric ceramics, adding sintering aids, such as low-melting glass or oxides, is the common method. But the microwave dielectric properties of the materials will be degraded seriously

*Corresponding author. Tel./fax: +86 77 3589 6290. E-mail address: fanglianggl001@yahoo.cn (L. Fang). as the glass or oxides will produce a large amount of amorphous phase in the materials.

Recently, some potential candidates for LTCC substrates have been concentrated on the compounds containing low melting point oxides [10–15]. Among these compounds, Mg₂V₂O₇ and R₂V₂O₇ (R=Ba, Sr, and Ca) ceramics possessed a good combination of microwave dielectric characteristics: $\varepsilon_r = 10 \sim 12.1$, $Q_u \times f = 15,200$ 58,300 GHz, and $\tau_f = -26 \sim 35 \text{ ppm/}^{\circ}\text{C}$ [14,15]. However, no investigations on their chemical compatibility with silver has been reported to date. Recently, we reported a low ε_r and high $Q_u \times f$ microwave dielectric ceramic LiCa₃MgV₃O₁₂ with cubic garnet-structure, which could be well sintered at 900 °C and was compatible with Ag electrode [16]. In Na₂O-CaO-MgO-V₂O₅ system, NaCa₂Mg₂V₃O₁₂ also adopts the cubic garnet structure reported by Bayer and was synthesized at 750 °C [17]. Therefore, the NaCa₂Mg₂V₃O₁₂ ceramic might have a low sintering temperature as well as good microwave dielectric properties to meet the requirement for LTCC application. In the present paper, the preparation and microwave dielectric properties of NaCa₂Mg₂V₃O₁₂ ceramic were reported for the first time, and its chemical compatibility with silver electrode was also investigated to check whether the NaCa₂Mg₂V₃O₁₂ ceramic is suitable for the LTCC technology. Adjusting the temperature coefficients of resonant frequency (τ_f) of NaCa₂Mg₂V₃O₁₂ ceramic by the addition of CaTiO₃ with a large positive τ_f value (~+800 ppm/°C)^[18] was also studied.

2. Experimental procedure

NaCa₂Mg₂V₃O₁₂ ceramics were prepared using the solidstate ceramic route from high-purity powder of Na₂CO₃, $CaCO_3$, MgO, and V_2O_5 (> 99%, Guo-Yao Co. Ltd., Shanghai, China). Since MgO is hygroscopic, it was calcined at 800 °C for 2 h to remove moisture retains. Powders were mixed and milled for 4 h using a planetary mill (Nanjing Machine Factory, Nanjing, China) operating at a running speed of 150 rpm with ZrO₂ media using alcohol as a medium. The mixtures were dried and calcined at 800 °C for 4 h. The calcined powder was re-milled for 6 h using 5 wt% of polyvinyl alcohol solution as a binder. The obtained powder was then crushed into a fine powder through a sieve with a 200 mesh. The obtained fine powder was then pressed into pellets of 12 mm in diameter and 7 mm in height by uniaxial pressing under a pressure of 200 MPa. The pellets were heated at 600 °C for 4 h to remove the PVA and then sintered on a Pt foil at 885–945 $^{\circ}$ C for 4 h.

The phase composition of the samples was analyzed by an X-ray diffraction measurement ($CuK\alpha_1$, 1.54059 Å, Model X'Pert PRO, PANalytical, Almelo, Holland) with $CuK\alpha$ radiation generated at 40 kV and 100 mA. The bulk densities of sintered ceramics were measured by Archimedes' method. The surface microstructure of the samples was examined using a field emission scanning electron microscope (FE-SEM, Model S4800, Hitachi, Japan). Composition analysis was performed using energy-dispersive spectroscopy (EDS, IE 350; INCA, Oxford, U.K.). The microwave dielectric properties were analyzed using a network analyzer (Model N5230A, Agilent Co., Palo Alto, Canada) and a temperature chamber (Delta 9039, Delta Design, San Diego, CA). The temperature coefficient of resonant frequency τ_f values were calculated using the following formula:

$$\tau_f = \frac{f_{85} - f_{25}}{60 f_{25}} \tag{1}$$

where f_{85} and f_{25} are the resonant frequencies at 85 °C and 25 °C, respectively.

3. Results and discussion

The X-ray diffraction pattern of the $NaCa_2Mg_2V_3O_{12}$ (NCMV) ceramic sintered at 915 °C for 4 h is shown in Fig. 1, and it can be fully indexed as single-phase with cubic garnet structure, which agreed well with the result reported by Bayer [17]. The FE-SEM micrograph of the surface of NCMV ceramic is also shown in Fig. 1. Homogenous microstructure with grains size in the range 1–5 μ m and few pores could be observed for NCMV ceramic sintered at 915 °C for 4 h.

The variations of relative density, ε_r , $Q_u \times f$, and τ_f of NaCa₂Mg₂V₃O₁₂ ceramics as a function of the sintering

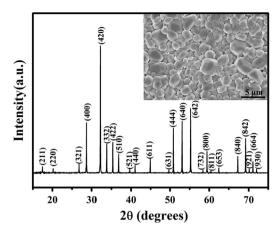


Fig. 1. X-ray diffraction result and FE-SEM image for the surface of a $NaCa_2Mg_2V_3O_{12}$ ceramic sintered at 915 °C for 4 h.

temperature were presented in Fig. 2. The relative density of NaCa₂Mg₂V₃O₁₂ ceramic increased with increasing sintering temperature, reached a maximum value (95.8%) at 915 °C, and then decreased slightly. Microwave dielectric properties versus sintering temperature of NCMV ceramics exhibited a trend similar to that of the relative density. ε_r increases from 8.8 to 10 as sintering temperature increases from 885 to 915 °C. Then ε_r reaches saturation and maintains stability, which indicates that the ε_r of ceramic is associated with the elimination of the pores [10,19]. The influence of the porosity on the microwave permittivity could be eliminated by applying Bosman and Havinga's correction [20] as shown in Eq. (2).

$$\varepsilon_{corrected} = \varepsilon_m (1 + 1.5p)$$
 (2)

where $\varepsilon_{corrected}$ and ε_{m} are corrected and measured values of permittivity, respectively. p is the fractional porosity (0.042 in this work). The porosity corrected permittivity is about 10.75. The $Q_{u} \times f$ value of NCMV ceramics reaches the maximum with a value of 50,600 GHz. Thereafter, the $Q_{u} \times f$ values decrease with further increasing temperature, which may be due to the decreased densification caused by the evaporation of Na₂O [21]. τ_{f} values of NCMV ceramic do not change remarkably with increasing sintering temperature and remain stable at about -47 ppm/°C.

NaCa₂Mg₂V₃O₁₂ ceramic developed in the present study has a low sintering temperature (915 °C), a good combination of microwave dielectric properties ($\varepsilon_r = 9.9$, $Q_u \times f = 50,600$ GHz), low bulk density ($< 3.5 \text{ g/cm}^3$) and low cost (cheap raw materials). However, the temperature stability of resonant frequency is poor $(\tau_f = -47 \text{ ppm/}^{\circ}\text{C})$ which precludes its use as substrate material for practical applications. The value of τ_f can be tuned nearly to zero by adding ceramics with a large positive τ_f value like CaTiO₃ ($\tau_f = +800 \text{ ppm/}^{\circ}\text{C}$) [18]. A similar approach was recently used by Huang and Tseng to tailor τ_f of Ca₄MgNb₂TiO₁₂ ceramics [22]. The addition of 30 mol% CaTiO₃ to Ca₄MgNb₂TiO₁₂ ($\tau_f \sim -32$ ppm/°C) results in a ceramic composite exhibiting $\varepsilon_r \sim 43.88$, $Q_u \times f \sim 20,200$ GHz and $\tau_f \sim -6.9 \text{ ppm/}^{\circ}\text{C}$ [23]. In this paper, the microwave dielectric properties and XRD patterns of (1-x) $NaCa_2Mg_2V_3O_{12}$ - $xCaTiO_3$ ($0 \le x \le 0.2$) ceramics sintered at

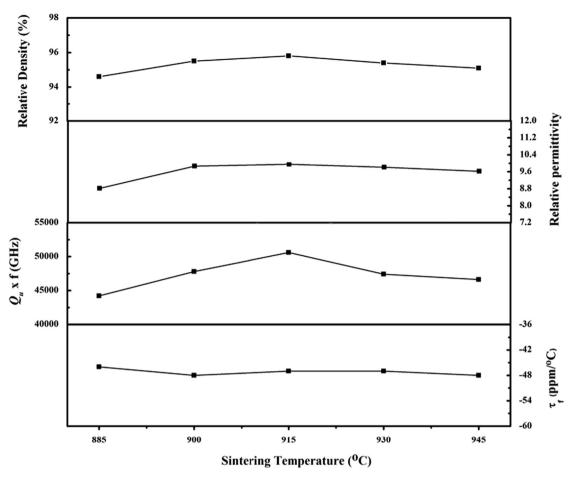


Fig. 2. The relative density, relative permittivity, quality factor, and temperature coefficient of the resonant frequency of $NaCa_2Mg_2V_3O_{12}$ ceramic as a function of sintering temperature.

Table 1
Bulk density and microwave dielectric properties of (1-x)NaCa₂Mg₂V₃O₁₂-xCaTiO₃ $(0 \le x \le 0.2)$ ceramics sintered at 930 °C for 4 h.

x value	Bulk density (g/cm ³)	$arepsilon_r$	$Q_u \times f$ (GHz)	$\tau_f (\text{ppm/}^{\circ}\text{C})$
0	3.28	9.8	47,400	-47
0.05	3.25	10.1	45,600	-40
0.1	3.20	10.7	41,900	-28
0.2	3.16	11.5	37,500	2

930 °C for 4 h are shown in Table 1 and Fig. 3. The ε_r values increased from 9.8 to 11.5 and the τ_f values increased from -47 to 2 ppm/°C with increasing x values from 0 to 0.6. All the sintered ceramics exhibited high $Q_u \times f$ values. The 0.8NaCa₂Mg₂V₃O₁₂-0.2CaTiO₃ ceramic sintered at 930 °C exhibits favorable microwave dielectric properties with ε_r =11.5, $Q_u \times f$ =37,500 GHz, and τ_f =2 ppm/°C. Fig. 4 shows the result from energy-dispersive spectroscopy of second phase in 0.8NaCa₂Mg₂V₃O₁₂-0.2CaTiO₃ ceramic sintered at 930 °C. Ca and Ti element are detected at a ratio of approximately 1:1, which confirms CaTiO₃ as the secondary phase.

For chemical compatibility experiment, mixtures of ceramic powders with 20 wt% Ag powders were cofired and analyzed to detect interactions between the low-fired samples and

electrodes. The X-ray powder diffraction pattern (XRPD) and the backscattered electron image of NaCa₂Mg₂V₃O₁₂ cofired with Ag at 915 °C are presented in Fig. 5. Since the XRD pattern does not show the formation of any other phase except for NaCa₂Mg₂V₃O₁₂ and Ag, and the backscattered electron image of NaCa₂Mg₂V₃O₁₂/Ag sample cofired at 915 °C for 2 h reveals no interaction to form new phases after firing, then it is considered that the NaCa₂Mg₂V₃O₁₂ compound has a chemical compatibility with silver.

4. Conclusions

Novel low firing temperature vanadate garnet NaCa₂Mg₂V₃O₁₂ ceramic was investigated as promising low-loss dielectric ceramic

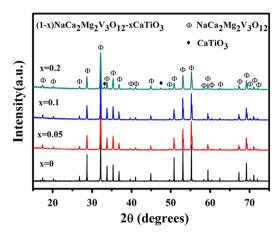


Fig. 3. XRD patterns of (1-x)NaCa₂Mg₂V₃O₁₂-xCaTiO₃ $(0 \le x \le 0.2)$ ceramics sintered at 930 °C for 4 h.

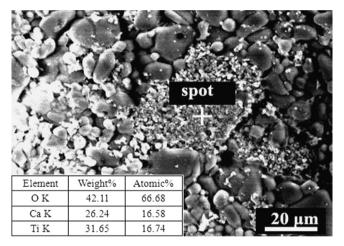


Fig. 4. FE-SEM image of $0.8NaCa_2Mg_2V_3O_{12}$ – $0.2CaTiO_3$ ceramic sintered at 930 °C. Inset shows result of energy-dispersive spectroscopy of second phase in $0.8NaCa_2Mg_2V_3O_{12}$ – $0.2CaTiO_3$.

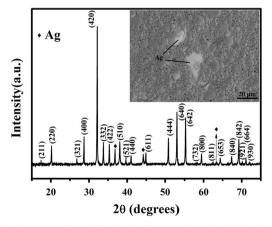


Fig. 5. XRD pattern of a $NaCa_2Mg_2V_3O_{12}$ ceramic mixed with 20 wt% Ag sintered at 915 °C for 2 h. Inset shows a backscattered electron image of $NaCa_2Mg_2V_3O_{12}$ sample cofired with Ag.

for microwave applications. NCMV ceramic can be densified at a relatively low sintering temperature of 915 °C and exhibited promising microwave dielectric properties of ε_r = 10, $Q_u \times f$ = 50,600 GHz, and τ_f = -47 ppm/°C. It was showed that

the microwave properties were improved by adding a small amount of CaTiO₃ into NCMV, and the $0.8 \text{NaCa}_2 \text{Mg}_2 \text{V}_3 \text{O}_{12}$ – 0.2CaTiO_3 ceramic sintered at $930\,^{\circ}\text{C}$ for 4 h exhibited improved properties of ε_r =11.5, $Q_u \times f$ =37,500 GHz, and τ_f =2 ppm/ $^{\circ}\text{C}$. The backscattered electron image and XRD analysis confirmed that the ceramic did not react with silver powders, indicating that NaCa₂Mg₂V₃O₁₂ might be an attractive promising candidate for LTCC application.

Acknowledgments

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