

Effects of annealing temperature on the microstructure, optical, ferroelectric and photovoltaic properties of BiFeO₃ thin films prepared by sol–gel method

Zebin Lin^a, Wei Cai^{a,*}, Weihai Jiang^a, Chunlin Fu^{a,b}, Chun Li^c, Yunxia Song^a

^a*School of Metallurgical and Materials Engineering, Chongqing University of Science and Technology, Huxi Town, Shapingba, Chongqing 401331, China*

^b*State Key Laboratory of Electronic Thin Films and Integrated Devices, Chengdu 610054, China*

^c*Chemical Engineering Institute, China University of Petroleum, Qingdao 266555, China*

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Abstract

Bismuth ferrite thin films were prepared via sol–gel spin-coating method and the effects of annealing temperature on microstructure, optical, ferroelectric and photovoltaic properties have been investigated. The results show that the bismuth ferrite thin films annealed at 550 °C is single phase and the grain size increases with the rise of annealing temperature. The band gap of bismuth ferrite thin films annealed at 550–650 °C is between 2.306 eV and 2.453 eV. With the rise of the annealing temperature, the remnant polarization gradually decreases and the coercive electric field increases. The short circuit photocurrent density decreases with the rise of annealing temperature, and the open circuit photovoltage and the power conversion efficiency of bismuth ferrite thin films annealed at 550 °C are higher than the thin films annealed at higher temperature. © 2013 Elsevier Ltd and Techna Group S.r.l. All rights reserved.

Keywords: A. Annealing temperature; C. Ferroelectric properties; C. Photovoltaic; D. Bismuth ferrite

1. Introduction

Over the past decades, many efforts have been made to develop renewable energy and clean fuel sources. In particular, solar energy has drawn great attention, in particular harvesting light energy through photovoltaic (PV) effect. Photovoltaic effect mainly involves three basic processes: (1) the generation of electron-hole pairs under illumination, (2) the separation of the electrons and holes, and (3) the movement of the electrons and holes. The charge separation in photovoltaic effect typically results from the following four mechanisms: (1) the poorly controlled granular interface [1], (2) the inherent non-centrosymmetry of bulk material [2], (3) the p–n junction or heterojunction (such as Schottky barrier) as in silicon-based solar cells [3,4], and (4) the built-in electric field caused by ferroelectricity or the domain walls [5,6]. It is noticeable that the ferroelectric thin films have a built-in electric field originating from polarization. Thus, PV effect in ferroelectric materials can be realized without p–n junction or Schottky barrier.

The PV effect of ferroelectric materials, such as BaTiO₃ [7], Pb(Zr,Ti)O₃ [8–14], (K_{0.5}Na_{0.5})(Mn_{0.005}Nb_{0.995})O₃ [15,16], (Bi_{3.7}Nd_{0.3})Ti₃O₁₂ [17] and BiFeO₃ [18–24] has become a hot spot because of its potential applications for the photovoltaic and optoelectronic devices. Among these ferroelectric materials, BiFeO₃ (short for BFO) has the vast application prospect due to lead free, large spontaneous polarization, the photovoltaic characteristics controlled by electric or magnetic field and the relatively narrower band gap [25,26]. Furthermore, BFO is the only single-phase multiferroic material at room temperature because of its ferroelectric Curie temperature (~1120 K) and antiferromagnetic Neel temperature (~640 K) above room temperature [27,28]. BFO thin films can be prepared by pulsed laser deposition (PLD) [29,30], sol–gel method [31,32], molecular beam epitaxy [33], chemical vapor deposition [34] and RF magnetron sputtering [35,36]. The sol–gel method is low cost and simple for preparation of BFO thin films. Although there are a few studies about the electric properties of BFO thin films prepared by the sol–gel method, the effects of annealing temperature on photovoltaic properties of BFO thin films have not been systematically studied. We prepared the BFO thin films by the sol–gel method and studied the effects of

*Corresponding author. Tel.: +86 23 65023479; fax: +86 23 65023706.

E-mail addresses: caiwei_cqu@163.com,
caiwei_cqu@yahoo.com.cn (W. Cai).

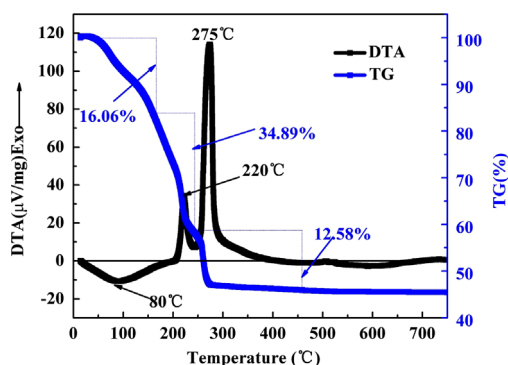


Fig. 1. DTA–TG curves of BFO gels.

annealing temperature on microstructure, optical, ferroelectric and photovoltaic properties and found out the best annealing temperature.

2. Experimental

2.1. Films preparation

BFO thin films were prepared by sol–gel method. Bismuth nitrate ($[\text{Bi}(\text{NO}_3)_3 \cdot 5\text{H}_2\text{O}]$, $\geq 99.0\%$, Sinopharm Group Co. Ltd.) and iron nitrate ($[\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}]$, $\geq 99.0\%$, Sinopharm Group Co. Ltd.) were used as raw materials. At the same time, acetic acid (CH_3COOH , short for HAc, $\geq 99.0\%$, Sinopharm Group Co. Ltd.) and 2-methoxyethanol ($\text{CH}_3\text{OCH}_2\text{OH}$, $\geq 99.0\%$, Sinopharm Group Co. Ltd.) were used as solvent and stabilizer, respectively. Firstly, $\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$ was dissolved in the mixed solvent of $\text{CH}_3\text{OCH}_2\text{OH}$ and HAc (the volume ratio of $\text{CH}_3\text{OCH}_2\text{OH}$ to HAc was 4:1) to form Fe precursor solution. The solution was continuously stirred at 50°C for 30 min to make sure $\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$ was completely dissolved. Then, Acetylacetone ($[\text{CH}_3\text{COCH}_2\text{COCH}_3]$, $\geq 99.0\%$, Sinopharm Group Co. Ltd.) was added to control the hydrolysis rate of $\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$. Dehydration of $\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$ was realized by adding acetic anhydride ($(\text{CH}_3\text{CO})_2\text{O}$, $\geq 99.0\%$, Chengdu Kelong Chemical Reagent Factory) to obtain a clear and homogenous solution. In the similar way, Bi precursor solution was prepared. Secondly, Fe precursor solution was added to Bi precursor solution by dropping in a molar ratio of Fe: Bi=1:1.05 (5 mol% excess Bi) to form the mixture, which was then stirred for 30 min at 25°C . Then, ethanolamine ($[\text{HOCH}_2\text{CHNH}_2]$, $\geq 99.0\%$, Chengdu Kelong Chemical Reagent Factory) was added to adjust the viscosity and pH of the solution. The concentration of the precursor was adjusted to 0.2 mol/L by adding proper amount of the mixed solvent. The obtained precursor solution was transparent and homogeneous. After aging the precursor solution for 24 h at 25°C , the thin films were deposited on the quartz glass and Pt(100 nm)/Ti(30 nm)/ SiO_2 (500 nm)/Si(500 μm) substrates by spin-coating method, respectively. Then, BFO thin films were annealed at 550 – 650°C for 10 min in oxygen atmosphere. The thickness of BFO thin films was controlled by repeating the spin coating and heat treatments.

2.2. Characterization

To find out appropriate annealing temperature for BFO thin films, thermal analysis of the xerogel was performed based on a thermoanalyzer (DTA–TG, JCR-2, Henven, China) in the temperature range from 0 to 750°C at a heating rate of $10^\circ\text{C}/\text{min}$. Al_2O_3 was used as reference material and sample holders. The phase analysis was carried out at room temperature by X-ray diffraction (DX-2700 model, Dandong Fangyuan, China, Cu target with a working voltage of 30 kV and a current of 20 mA) in a scanning rate of $3^\circ/\text{min}$. Surface morphology was determined by scanning electron microscope (S-3700N model, Hitachi, Tokyo, Japan). The thickness of BFO thin films is about 240 nm determined by step profiler (Dekatk 150, Veeco, USA). The optical transmission characteristics were studied by a double beam ultraviolet–visible (UV–vis) spectrophotometer (TU1810, Persee, China) in the wavelength range of 200–1100 nm. To measure the photovoltaic characteristic and electric properties, the Au top electrodes of 200 μm in a diameter were deposited on the BFO thin films using DC magnetron sputtering (JGP280, SKY, China). The ferroelectric and leakage measurements were made by ferroelectric test system (TF2000e, aixACCT, Germany). The current–voltage (I – V) behavior was characterized by a current source meter (2400, Keithley, USA) when the thin films were illuminated with the solar simulator (Newport, Oriel instrument, USA).

3. Results and discussion

3.1. Annealing temperature

Fig. 1 shows DTA–TG curves of BFO gels. Firstly, one endothermic reaction occurs near 80°C due to dehydration of residual water and removal of organic solvents, leading to a weight loss around 16.06%. Secondly, there are two obvious exothermic peaks near 220°C and 275°C , respectively. Between 180°C and 400°C , the significant weight loss around 47.47% (34.89%+12.58%) corresponds to pyrolysis of organic residues. Thirdly, there is a small endothermic peak near 593°C and the weight is unchanged, which implies that there is crystallization of BFO thin films. Therefore, the annealing temperature of BFO thin films is set as 550°C , 600°C and 650°C , respectively. The effects of annealing temperature on microstructure, optical, ferroelectric and photovoltaic properties of BFO thin films have been investigated.

3.2. Microstructure

Fig. 2 shows the XRD patterns of BFO thin films annealed at different temperatures. Firstly, it is found that the main diffraction peaks in samples annealed at 550 – 650°C are indexed as a rhombohedral distortion perovskite structure (BiFeO_3 , $R3m$ space group, JCPDS card no. 73–2548). No secondary phase is observed in the sample annealed at 550°C . But there is secondary phase identified as $\text{Bi}_2\text{Fe}_4\text{O}_9$ (JCPDS card no. 74-1098 marked with # in Fig. 2) in the samples

annealed at 600 °C and 650 °C, which is commonly observed in this system [24,25]. It is worthwhile to note that the intensity of diffraction peaks corresponding to the secondary phase becomes stronger with the rise of annealing temperature. The content of $\text{Bi}_2\text{Fe}_4\text{O}_9$ in the sample annealed at 650 °C reaches 34.26 wt% by means of XRD quantitative analysis. The emergence of secondary phase $\text{Bi}_2\text{Fe}_4\text{O}_9$ is attributed to that the kinetics of phase formation because of the volatilization of some reactants and phase decompositions at high temperature [37].

The grain size of BFO thin films was determined by Scherer's formula:

$$d = \frac{K\lambda}{\beta \cos \theta} \quad (1)$$

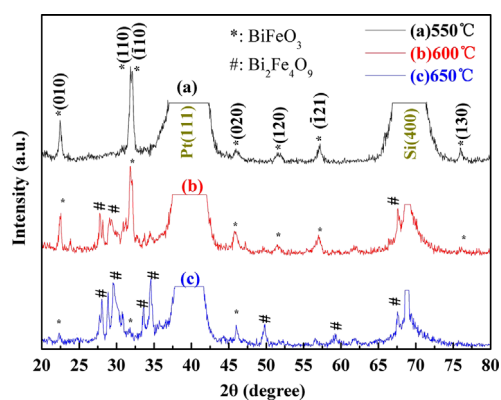


Fig. 2. XRD patterns of BFO thin films annealed at different temperatures.

where d is the grain size (nm), K is the Scherrer constant (0.89), λ is the wavelength of X-rays (0.15406 nm), β is the full width at half maximum reflection height and θ is the diffraction angle. Based on Eq. (1), the grain size of BFO thin film annealed at 550 °C, 600 °C and 650 °C is 20 nm, 22 nm and 29 nm, respectively. It indicates that the grain size increases with the rise of annealing temperature.

Fig. 3 shows SEM micrographs of BFO thin films annealed at different temperatures. Firstly, it is found that the BFO thin films annealed at 550–650 °C are crack-free, homogeneous and dense. The BFO thin films annealed at 650 °C is denser than that annealed at 550 °C and 600 °C. It may be due to the higher crystallinity in the sample annealed at 650 °C. Secondly, the grain size increases with the rise of annealing temperature. In particular, the grain size of the sample annealed at 650 °C is obviously larger than that annealed at 550 °C and 600 °C, which is consistent with the calculated results according to the Scherrer formula. Moreover, there are some obvious white particulates in the sample annealed at 650 °C. According to the XRD patterns, it is inferred that the white particulates is secondary phase $\text{Bi}_2\text{Fe}_4\text{O}_9$.

3.3. Optical properties

Fig. 4(a) shows the optical transmission spectra and absorption spectra of BFO thin films annealed at different temperatures. The absorption coefficient α is deduced by the formula:

$$\alpha = 1/t \ln(1/T) \quad (2)$$

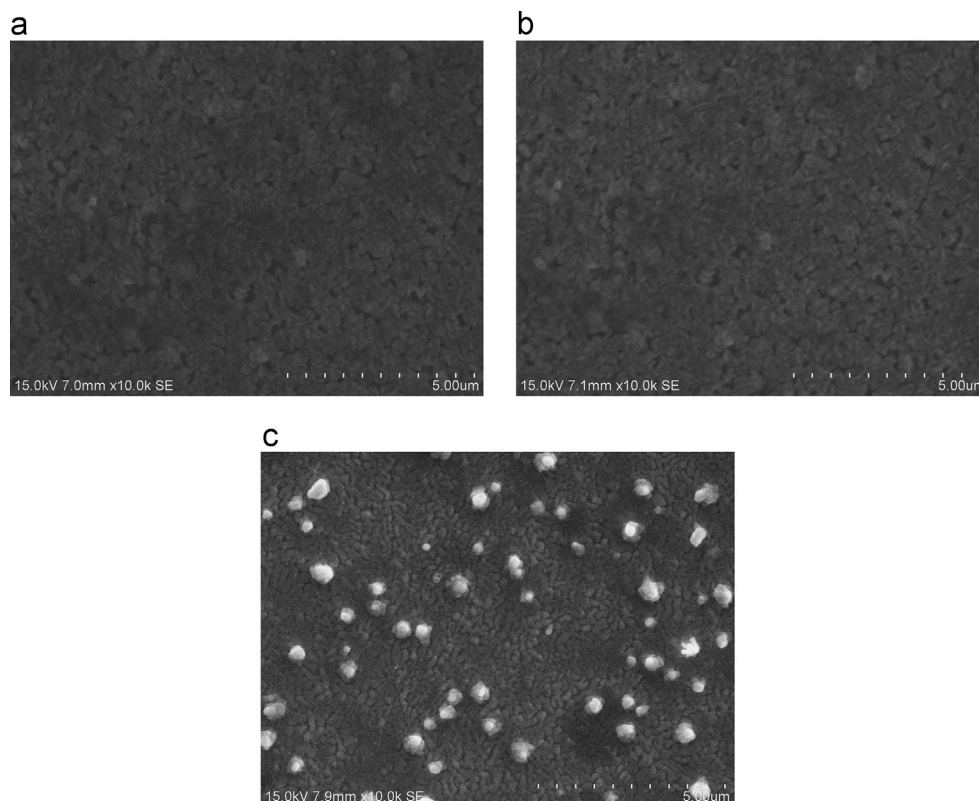


Fig. 3. SEM micrographs of BFO thin films annealed at different temperatures (a) 550 °C, (b) 600 °C, and (c) 650 °C.

where t means the thickness of the film (t of BFO thin films has been determined by step profiler) and T represents the transmittance. The region of the absorption edge shows that the fundamental absorption edge for these samples is around 1.62 eV (corresponding to 766 nm), which means the BFO thin films can absorb visible light, while typical ferroelectric materials such as BaTiO₃-based thin films only has a good absorption of ultraviolet light because of its wide band gap (> 3 eV) [38]. Although the overall absorption profiles for BFO thin films annealed at 550–650 °C are similar, there are some differences in the higher absorption region (energy is between 2.5 eV and 6.25 eV). The absorption coefficient α of the sample annealed at 650 °C is obviously lower than that annealed at 550 °C and 600 °C. It may be caused by the impurity phase resulting from the higher annealing temperature. The result of transmission spectra is in agreement with that of absorption spectra.

The optical band gap of BFO thin films was calculated via Tauc's Law

$$(\alpha h\nu)^n = A(h\nu - E_g) \quad (3)$$

where A is a constant, α is the absorption coefficient, $h\nu$ is the photon energy, n represents the type of optical transition between valence and conduction bands, E_g is the optical band gap. For $n=2$, the transition is direct, while $n=0.5$ indicates an indirect optical transition [39]. However, there is the dispute about n value of BFO thin films. Ihlefeld et al. [40] et al. found that the BFO thin films is direct band gap semiconductor, n is 2. But some researchers contend that the BFO thin films is indirect band gap semiconductor, n is 0.5 [41,42]. To solve the problem, the curves were obtained by plotting $(\alpha h\nu)^2$ and versus $h\nu$, $(\alpha h\nu)^{0.5}$ and versus $h\nu$, respectively (shown in the inner figure of Fig. 4(b)). In our data, the lack of the characteristic shape of the $(\alpha h\nu)^{0.5}$ versus $h\nu$ plot indicates the required phonon participation argues against an indirect gap. Therefore, E_g can be obtained by plotting $(\alpha h\nu)^2$ versus $h\nu$ and extrapolating the linear portion of the plot to $(\alpha h\nu)^2=0$ (shown in Fig. 4(b)). The band gap values of the samples annealed at different temperatures are listed in the table of Fig. 4(b). These values are similar to Ihlefeld's result [40]. The band gap of the samples is between 2.306 eV and 2.453 eV, which indicates that the BFO thin films can absorb visible light. The band gap of BFO thin films annealed at 600 °C and 650 °C is lower than that of BFO thin films annealed at 550 °C, which may be attributed to the improvement of crystallinity. The result is consistent with the Sharma's results [43,44]. But it is important to note that the band gap of BFO thin films annealed at 650 °C is slightly higher than that of the sample annealed at 600 °C. It may be due to the greater proportion of secondary phase Bi₂Fe₄O₉ in the sample annealed at 650 °C.

3.4. Ferroelectric properties

Fig. 5 shows the room temperature hysteresis loops of BFO thin films annealed at different temperatures measured at 200 Hz. Firstly, it is found that BFO thin films exhibit typical

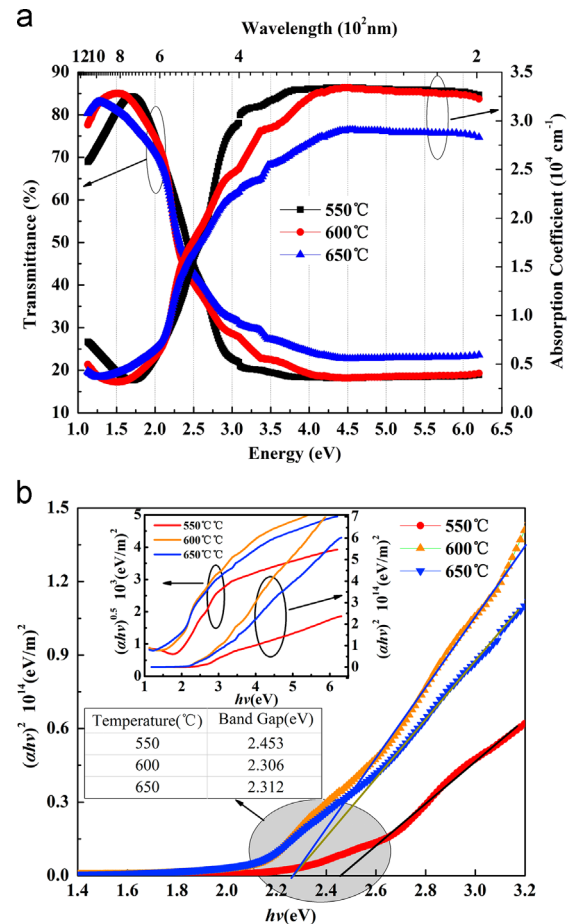


Fig. 4. (a) Transmission and absorption spectra and (b) dependence of $(\alpha h\nu)^2$ on $h\nu$ of BFO thin films annealed at different temperatures.

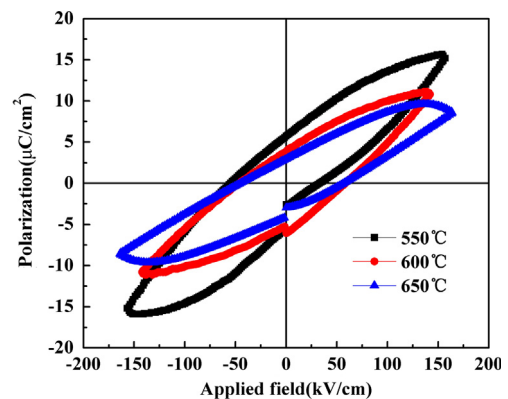


Fig. 5. Room temperature hysteresis loops of BFO thin film annealed at different temperatures.

hysteresis loops. The remnant polarization (P_r) and coercive electric field (E_C) of the samples annealed at different temperatures are shown in Table 1. As the annealing temperature rises, the remnant polarization gradually decreases and the coercive electric field increases. The change of the remnant polarization is due to two reasons. On the one hand, the secondary phase Bi₂Fe₄O₉, whose Curie temperature is −23 °C [45], in BFO thin films annealed at higher temperature is

Table 1

The remnant polarization and coercive electric field of BFO thin films annealed at different temperatures.

Annealing temperature (°C)	550	600	650
$2P_r$ ($\mu\text{C}/\text{cm}^2$)	11.44	7.74	5.91
$2E_c$ (kV/cm)	86.3	105.1	105.1

paraelectric phase at room temperature. It results in the decrease of remnant polarization. On the other hand, the higher oxygen vacancy concentrations result from more Fe^{2+} ions in the sample at higher annealing temperatures [28]. Oxygen vacancies act as space charge, which causes strong domain pinning. The higher annealing temperature leads to more oxygen vacancies. The increase of defect concentration results in the decrease in the remnant polarization [46]. In this manner, the remnant polarization of BFO thin films decreases with the rise of annealing temperature. Moreover, the increase of the coercive electric field in the sample annealed at higher temperature may be determined by the competition of the higher oxygen vacancy concentration and the larger grain size. On the one hand, oxygen vacancy may affect domain wall motion by screening of the polarization charge. A formation of mechanical barriers against the domain walls by oxygen vacancies, so called domain wall pinning, might also stabilize the domain configuration [47]. As mentioned above, high annealing temperature leads to the increase of oxygen vacancy concentration. It indicates that as annealing temperature rises, domain wall pinning enhances so that domain wall moves difficultly so as to increase the coercive electric field. On the other hand, barrier for switching ferroelectric domain must be broken through and energy barrier decreases as grain size increases. So the reversal polarization of a ferroelectric domain in a small grain is more difficult than that in a large grain [48]. As mentioned above, the grain size of BFO thin films increases with the rise of annealing temperature and leads to the decrease of coercive electric field. But in two opposite effects, the increase of oxygen vacancy concentration caused by higher annealing temperature is dominant, so the coercive electric field increases.

Fig. 6 shows DC leakage characteristic of BFO thin film annealed at different temperatures measured at 200 Hz and room temperature. It is found that the leakage current of BFO thin films is very sensitive to the annealing temperature, i.e. the higher the annealing temperature, the greater the leakage current. A similar dependence of leakage current on annealing temperature has been reported in BFO, La-doped BFO, Ce-doped BFO and Mn-doped BFO thin films [49–52]. The leakage behavior in BFO thin films depends on phase purity, grain size, defects such as oxygen vacancy, and so on. Among all the factors, oxygen vacancy may play an important role. Oxygen vacancies generate according to the reaction:



alternatively, reduction of Fe^{3+} to Fe^{2+} can be ruled out to compensate the space charges according to the reaction:

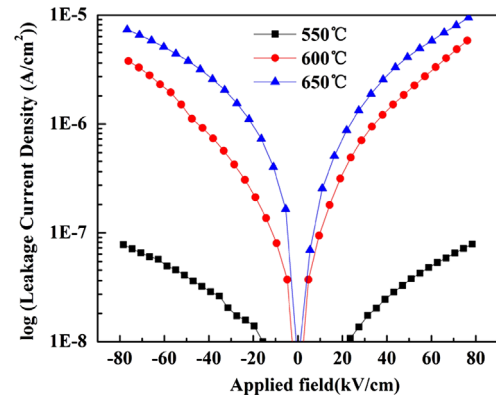
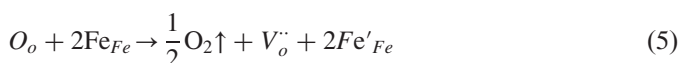


Fig. 6. DC leakage of BFO thin film annealed at different temperatures.

The oxygen vacancies formed during growth cause a portion of the Fe^{3+} ions to become Fe^{2+} . These Fe ions are responsible for the high leakage of BFO. The higher annealing temperature leads to the more oxygen vacancies, which causes the higher leakage [53,54].

3.5. Photovoltaic characteristic

Fig. 7 shows illuminated J – V curves and the terminal voltage dependences of power conversion efficiency for BFO thin films annealed at different temperatures. Firstly, it can be seen that the short circuit photocurrent density (the intercept of J -axis and J – V curve, short for J_{SC}) and the open circuit photovoltage (the intercept of V -axis and J – V curve, short for V_{OC}) are negative. This is due to the different work functions between BFO thin films and the electrodes [55]. Secondly, the short circuit photocurrent density and open circuit photovoltage of BFO thin films are shown in Table 2. It is found that the short circuit photocurrent density decreases with the rise of annealing temperature, and the open circuit photovoltage of the sample annealed at 550 °C is higher than that of the samples annealed at higher temperature. It is well known that the J_{SC} and V_{OC} of ferroelectric thin films under illumination depend heavily on the built-in electric field caused by the remnant polarization. As mentioned above, the remnant polarization of BFO thin films decreases with the rise of annealing temperature. Therefore, the higher photovoltaic output in BFO thin films annealed at 550 °C is due to its higher internal depolarization field caused by the larger remnant polarization. The result is similar to Qin's [55] and Zheng's [56] results. It is believed that the larger depolarization field could separate the electron and holes more effectively and enhance the photovoltaic output of BFO thin films. Thirdly, the light-to-electricity power conversion efficiency η can be determined from the illuminated J – V curves (shown in Fig. 7(b)). The input power density P_{in} (mW/cm^2) is the simulated solar light intensity I_{so} (mW/cm^2), and the photovoltaic output power density is $P_{out} = JV$ (mW/cm^2). The power conversion efficiency η at the point (J , V) can be calculated as:

$$\eta = \frac{P_{out}}{P_{in}} = \frac{P_{out}}{I_{so}} = \frac{JV}{I_{so}} \quad (6)$$

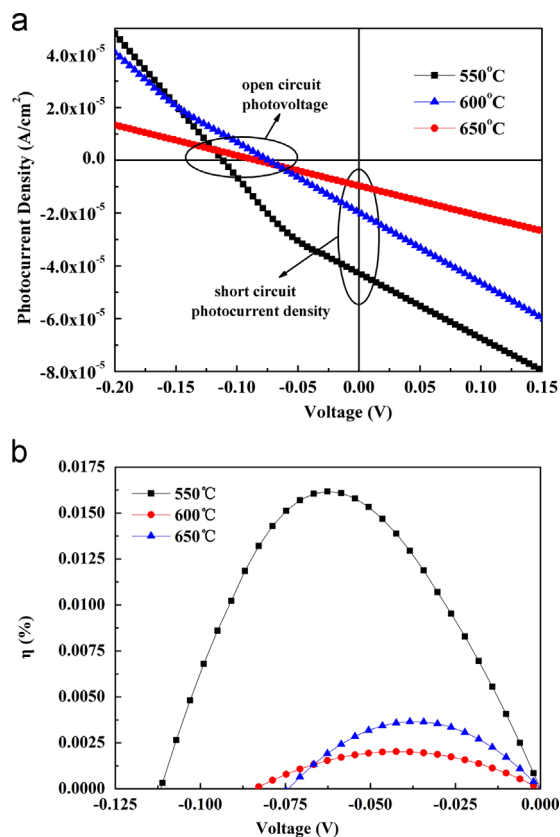


Fig. 7. (a) Illuminated $J-V$ curves and (b) corresponding terminal voltage dependence of light-to-electricity power conversion efficiency for BFO thin films annealed at different temperatures.

Table 2

The short circuit photocurrent density and the open circuit photovoltage of BFO thin films annealed at different temperatures.

Annealing temperature (°C)	550	600	650
J_{SC} (10^{-5} A/cm ²)	4.327	2.031	0.990
V_{OC} (V)	0.11	0.075	0.078

based on the illuminated $J-V$ curve, the voltage dependence of power conversion efficiency of BFO thin films is presented in Fig. 7(b). It can be found that the maximum power conversion efficiency η_{max} of the samples annealed at 550–650 °C occurs at about half of the open circuit photovoltage. The η_{max} of BFO thin films annealed at 550 °C, 600 °C and 650 °C is 0.0161%, 0.00203%, and 0.00366%, respectively. The power conversion efficiency is related to the separation and movement of carriers driven by the polarization. The higher remnant polarization of the samples annealed at 550 °C provides the larger electric field as driving force to separate the light electron-hole carriers and leads to higher power conversion efficiency. But it is worthwhile to note that although the remnant polarization of BFO thin films annealed at 600 °C is higher than that of the sample annealed at 650 °C, the power conversion efficiency of thin films annealed at 600 °C is lower than that of the sample annealed at 650 °C. It may be related to

more secondary phase in BFO thin films annealed at 650 °C. This problem will be further studied. The power conversion efficiency of BFO is higher than that of wide band gap PZT-based thin films [57], but is lower than that of epitaxial BFO thin films [58]. Moreover, the power conversion efficiency of Au/BFO/Pt structure in our paper is lower than that of ITO/BFO/Pt, which can absorb more visible light [59]. In summary, the single-phase BFO thin films annealed at 550 °C has narrow band gap, relatively lower leakage current, larger remnant polarization, lower coercive electric field, larger photovoltaic output and light-to-electricity power conversion efficiency. Therefore, 550 °C for BFO thin films prepared by sol-gel method is the best annealing temperature.

4. Conclusions

BFO thin films were prepared by sol-gel method and the effects of annealing temperature on microstructure, optical, ferroelectric and photovoltaic properties have been investigated. The BFO thin films annealed at 550 °C is single phase and the higher annealing temperature leads to the existence of secondary phase $Bi_2Fe_4O_9$. The grain size increases with the rise of annealing temperature. BFO thin films annealed at 550–650 °C can absorb visible light because of its narrow band gap (2.306–2.453 eV). As the annealing temperature rises, the remnant polarization gradually decreases and the coercive electric field increases. The short circuit photocurrent density of BFO thin films decreases with the rise of annealing temperature, and the open circuit photovoltage and power conversion efficiency of BFO thin films annealed at 550 °C is higher than that of the sample annealed at higher temperature. It results from the change of remnant polarization.

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