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## Short communication

# Effect of pressure loading cycle on spark plasma sintered ZrB<sub>2</sub>–SiC–Yb<sub>2</sub>O<sub>3</sub> ceramics

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#### Abstract

The influence of the pressure loading cycle on spark plasma sintered  $ZrB_2$ –SiC– $Yb_2O_3$  ceramics was investigated. Spark plasma sintering consists of three stages, i.e., a heating-up, isothermal and cooling stage. Applying the load during isothermal stage allowed evaporation of volatile species. Applying the load during the heating-up and isothermal stages resulted in the entrapment of volatile species and the formation of nanometer-sized spherical  $SiO_2$  particles. The presence of  $SiO_2$  particles led to a large thermal residual stress, and improved fracture toughness in the  $ZrB_2$ –SiC– $Yb_2O_3$  ceramics.

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### 1. Introduction

ZrB<sub>2</sub>-based ceramics are especially promising for high-temperature structural applications, such as cutting tools, thermal protection for hypersonic vehicles, molten metal crucibles, etc. owing to their unique properties such as high melting temperature, high strength, high thermal and electrical conductivity, and excellent resistance to thermal shock and oxidation [1,2]. ZrB<sub>2</sub>-based ceramics have typically been densified by hot pressing (HP) at high temperature with a 1–2 h holding time, which in turn leads to grain growth and thus limited strength. In comparison with conventional HP, spark plasma sintering (SPS) allows to densify ZrB<sub>2</sub>-based ceramics within minutes applying a very fast heating and cooling rate cycle [2–10].

To further adjust the microstructure and improve the mechanical properties of SPSed ZrB<sub>2</sub>-based ceramics, the effect of various sintering parameters including temperature, holding time, heating rate, and pressure were conducted. Guo et al. systematically studied the effect of heating rate, soaking

Typically, spark plasma sintering comprises three stages, i.e. a fast heating-up stage, a short isothermal dwell stage, and a rapid cooling stage. The mechanical pressure can be applied at any moment during the thermal cycle. The influence of the pressure loading cycle on the microstructure and fracture toughness of SPS ZrB<sub>2</sub>–SiC–Yb<sub>2</sub>O<sub>3</sub> is investigated.

## 2. Experimental procedure

The raw materials used were  $ZrB_2$  (Grade B, 2  $\mu m,$  H.C. Starck, Karlsruhe, Germany), SiC (Grade HSC-059, 0.7  $\mu m,$ 

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time, and sintering temperature on the microstructure of SPS ZrB<sub>2</sub> ceramics, revealing that a rapid grain growth was observed at high sintering temperature, slow heating rates, and long holding times [6]. The density of SPS ZrB<sub>2</sub>–SiC composites was reported to increase with increased holding time and pressure, and prolonging the holding time could improve the fracture toughness [8]. We recently reported on the influence of the heating rate on densification, microstructure, and strength of SPS ZrB<sub>2</sub>–SiC composites with or without Yb<sub>2</sub>O<sub>3</sub> addition, showing that a low heating rate induced grain growth and slightly increased the strength of the Yb<sub>2</sub>O<sub>3</sub>-free ceramics, whereas the grain size was hardly influenced and the strength decreased in Yb<sub>2</sub>O<sub>3</sub> doped composites [9].

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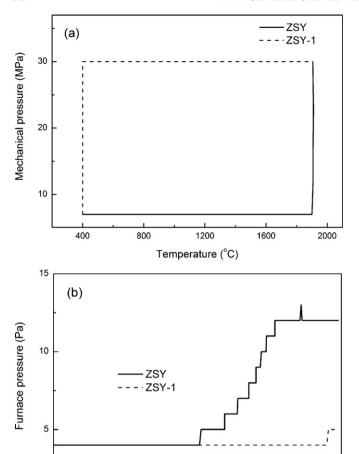


Fig. 1. Mechanical load (a) and vacuum evolution (b) in the furnace as a function of temperature during the SPS process.

1600

Temperature (°C)

1800

Superior Graphite, USA), Yb<sub>2</sub>O<sub>3</sub> (99.9%, ChemPur, Germany), and analytical grade nitric acid (65%). The starting powder composition of the investigated ZrB<sub>2</sub>–SiC–Yb<sub>2</sub>O<sub>3</sub> ceramics contained 75 vol% ZrB<sub>2</sub>, 20 vol%SiC and 5 vol% Yb<sub>2</sub>O<sub>3</sub>. A colloidal suspension coating technique was used to incorporate

the Yb<sub>2</sub>O<sub>3</sub>. The same coating technique was successfully applied to produce rare earth oxide doped ZrO<sub>2</sub> powders for the synthesis of tetragonal zirconia polycrystalline ceramics [11]. First, the appropriate amount of Yb2O3 was dissolved in nitric acid to obtain a clear nitrate solution. Subsequently, the solution was mixed with ZrB<sub>2</sub> and SiC powders in a polyethylene jar using ethanol and ZrO2 milling balls (grade TZ-3Y, Tosoh, Tokyo, Japan) for 24 h, followed by rotating evaporation to remove the solvent. After drying, the powder mixture was put into a graphite die/punch set-up lined with graphite paper and densified by SPS (Type HP D 25/1, FCT Systeme, Rauenstein, Germany). Details on the experimental SPS set-up are provided elsewhere [12]. SPS was performed at 1900 °C for 4 min in vacuum with a heating rate of 100 °C/min. Fig. 1(a) shows the mechanical load during the SPS process. In the ZSY ceramics, the initial mechanical load of 7 MPa was increased to 30 MPa within 1 min upon reaching 1900 °C. In the ZSY-1 ceramics, the pressure was already increased from 7 to 30 MPa within 1 min at 400 °C. The pressure was released at the onset of cooling.

The microstructure was observed by scanning electron microscopy (SEM, XL-30FEG, FEI, Eindhoven, The Netherlands) equipped with backscattered electron imaging (BSE), and transmission electron microscopy (TEM, JEM2100, JEOL, Japan). The fracture toughness was measured by the Vickers indentation method (Model FV-700, Future-Tech Corp., Tokyo, Japan) using a load of 98 N for 10 s on a polished surface.

#### 3. Results and discussion

Backscattered electron images of polished surfaces of ZSY and ZSY-1 are shown in Fig. 2. The ZSY ceramics contain a grey  $ZrB_2$  matrix with a dispersed dark SiC phase and an additional white contrast phase, as shown in Fig. 2(a). We have previously reported the XRD results of ZSY ceramics [9,10]. Based on XRD analysis, the white phase was identified as  $Yb_2Zr_8O_{19}$  (ICDD 78-1309) and  $YbBO_3$  (ICDD 19-1427), formed by the reaction of  $ZrO_2$  and  $B_2O_3$  with  $Yb_2O_3$  [9,10]. Because of the small difference in mean atomic number, the  $Yb_2Zr_8O_{19}$  and  $YbBO_3$  phases cannot be differentiated by BSE contrast. Although the XRD patterns of the ZSY-1 and ZSY

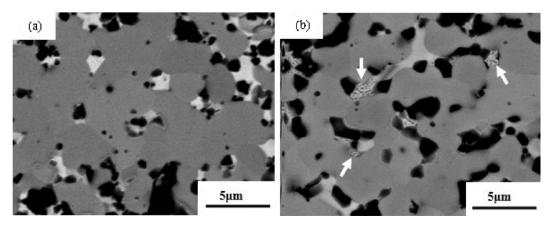


Fig. 2. Backscattered electron images of polished surfaces of ZSY (a) and ZSY-1 (b).

ceramics completely coincided, an additional white phase with darker dispersed features is present in the microstructure of ZSY-1, as indicated by the white arrows in Fig. 2(b). Transmission electron microscopy (TEM) images of the grain boundary composite structure are shown in Fig. 3. The TEM image in combination with EDX point analysis revealed that the black spots were 100–200 nm sized spherical SiO<sub>2</sub> particles. The SiO<sub>2</sub> particles were mostly present in the multiple grain junctions.

It is well-known that  $SiO_2$  and  $B_2O_3$  are present on the surface of SiC and  $ZrB_2$  powders, respectively. During thermal treatment, the  $B_2O_3$  and  $SiO_2$  dissolve forming borosilicate glass ( $SiO_2$ – $B_2O_3$ ) in  $ZrB_2$ –SiC ceramics [13]. XRD analysis of  $ZrB_2$ –SiC– $Yb_2O_3$  ceramics revealed that the  $Yb_2O_3$  can react with  $B_2O_3$  from the borosilicate glass to form  $YbBO_3$  [9–10]. Due to the consumption of  $B_2O_3$ , the borosilicate glass changed into  $SiO_2$ . Under vacuum and high temperature conditions,  $SiO_2$  can be reduced by SiC as follows [14]:

$$2SiO_2 + SiC \rightarrow 3SiO(g) + CO(g)$$
 (1)

In an open porous ceramic system, such as during heating under low load conditions like for ZSY, SiO vapour could readily be removed under the applied vacuum conditions, which would even further promote reaction (1) to proceed to completion. This would explain the fact that SiO<sub>2</sub> particles were not observed in the ZSY ceramics. When the pressure, however, was pre-applied during the heating-up stage, the escape of gaseous SiO was limited or even inhibited resulting in the precipitation of SiO<sub>2</sub> in the ZSY-1 ceramics. The evaporation of volatile species during SPS of ZSY and the absence of that during SPS of ZSY-1 are proven by the vacuum level evolution in the furnace as a function of temperature, as shown in Fig. 1(b). The furnace pressure during SPS of ZSY clearly increased above about 1600 °C, whereas the vacuum level remained constant during SPS of ZSY-1.

The fracture toughness of ZSY-1,  $4.3 \pm 0.1$  MPa m<sup>1/2</sup>, was approximately 21% higher than for ZSY,  $3.5 \pm 0.1$  MPa m<sup>1/2</sup>. To elucidate the toughening mechanisms induced by changing the pressure loading cycle, fracture surfaces of ZSY and ZSY-1 were investigated by SEM, as shown in Fig. 4. The microstructure of both ceramic grades consisted of equiaxed ZrB<sub>2</sub> and SiC grains and the fracture surfaces revealed a transgranular fracture. Although the coefficient of thermal expansion (CTE) of Yb<sub>2</sub>Zr<sub>8</sub>O<sub>19</sub> and YbBO<sub>3</sub> phases are not known, the CTE of ZrB<sub>2</sub>  $(6.5 \times 10^{-6} \, ^{\circ}\text{C}^{-1})$  and SiC  $(4.7 \times 10^{-6} \, ^{\circ}\text{C}^{-1})$  are substantially higher than for SiO<sub>2</sub>  $(0.5 \times 10^{-6} \, ^{\circ}\text{C}^{-1})$ . Due to the large thermal expansion mismatch, very rapid cooling during SPS could introduce a

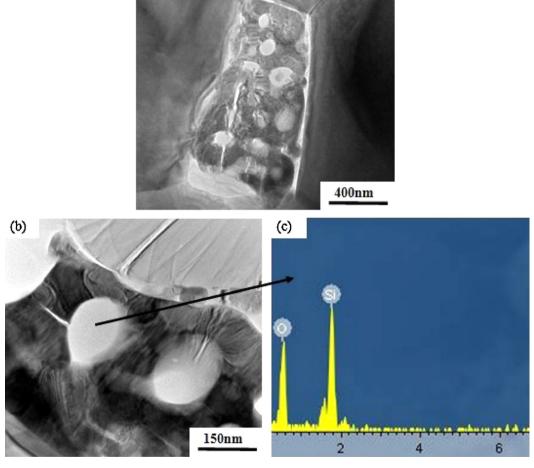


Fig. 3. Low (a) and high (b) magnification TEM images and EDS point analysis (c) of the spherical phase in the ZSY-1 ceramics.

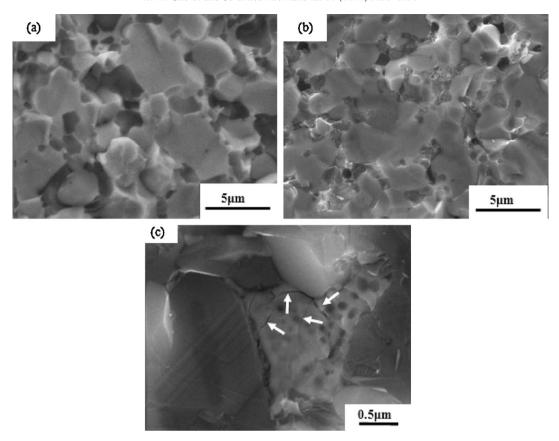


Fig. 4. SEM images of fracture surfaces of ZSY (a), and ZSY-1 (b and c) ceramics.

generation of large residual thermal stress in ZSY-1 ceramics. According to Selsing's model of a single spherical inclusion imbedded within an infinite isotropic elastic matrix, there would be thermal stress *P* presented on the spherical inclusion [15]:

$$P = \frac{\Delta \alpha \Delta T}{[(1+v_m)/2E_m + (1-2v_p)/E_p]}$$

where the subscripts m and p refer to the matrix and the particle, respectively,  $\Delta \alpha = \alpha_p - \alpha_m$  is the thermal expansion mismatch,  $\Delta T$  is the difference between the processing temperature and the room temperature, E is the elastic modulus, and v is the Poisson's ratio. In the ZSY-1 ceramics, the ZrB<sub>2</sub> was referred to as the matrix and the SiO<sub>2</sub> was referred to as inclusion. Based on Selsing's model, the estimated residual stress presented on  $SiO_2$  was about -1.1 GPa. Therefore, the  $SiO_2$ -containing glassy phase pocket is under compressive stress in ZSY-1 ceramics. Due to the presence of SiO<sub>2</sub> particles, the residual thermal stress of ZSY-1 ceramics was larger than that of ZSY ceramics. Monteverde has reported that the freezing of thermal strained field in a particulate-reinforced ceramic-matrix composite had a beneficial effect on fracture toughness [16]. The large thermal residual stress in the ZSY-1 ceramics led to an increased fracture toughness, compared to the ZSY ceramics. So, applying the mechanical pressure already during the heating-up stage during SPS led to an increased fracture toughness of the ZrB<sub>2</sub>–SiC–Yb<sub>2</sub>O<sub>3</sub> ceramics. Careful observation reveals the presence of microcracks in the ZSY-1 ceramics, as indicated by the white arrows in Fig. 4(c). The microcracks could result from the mechanical testing process due to the brittle feature of the SiO<sub>2</sub>-containing pocket.

## 4. Conclusion

The influence of the pressure loading cycle on ZrB<sub>2</sub>–SiC–Yb<sub>2</sub>O<sub>3</sub> ceramics was assessed. Microstructural observation revealed that nanometer-sized spherical SiO<sub>2</sub> particles are formed at the grain boundaries when already applying the load during the heating-up stage due to the entrapment of volatile species. The presence of SiO<sub>2</sub> particles led to a large thermal residual stress, and improved fracture toughness in the ZrB<sub>2</sub>–SiC–Yb<sub>2</sub>O<sub>3</sub> ceramics. Applying the load upon reaching the targeted SPS temperature allows evaporation of volatile species, avoiding the formation of SiO<sub>2</sub> precipitates and microcracks.

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