

Lead-free $(\text{Ba}_{0.85}\text{Ca}_{0.15})(\text{Ti}_{0.9}\text{Zr}_{0.1})\text{O}_3\text{--CeO}_2$ ceramics with high piezoelectric coefficient obtained by low-temperature sintering

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Abstract

Lead-free $(\text{Ba}_{0.85}\text{Ca}_{0.15})(\text{Ti}_{0.9}\text{Zr}_{0.1})\text{O}_3\text{--}x\%\text{CeO}_2$ (BCZT– $x\text{Ce}$) piezoelectric ceramics have been prepared by the traditional ceramic process and the effects of CeO_2 addition on their phase structure and piezoelectric properties have been studied. The addition of CeO_2 significantly improves the sinterability of BCZT ceramics which results in a reduction of sintering temperature from 1540 °C to 1350 °C without sacrificing the high piezoelectric properties. X-ray diffraction data show that CeO_2 diffuses into the lattice of BCZT and a pure perovskite phase is formed. SEM images indicate that a small addition of CeO_2 greatly affects the microstructure. Main piezoelectric parameters are optimized at around $x = 0.04$ wt% with a high piezoelectric coefficient ($d_{33} = 600$ pC/N), a planar electromechanical coefficient ($k_p = 51\%$), a high dielectric constant ($\epsilon_r = 4843$) and a low dissipation factor ($\tan \delta = 0.012$) at 1 kHz, which indicates that the BCZT– $x\text{Ce}$ ceramics are promising for lead-free practical applications.

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1. Introduction

It is well known that lead zirconate titanate (PZT) ceramics have widely been used for piezoelectric actuators, sensors and transducers due to their excellent piezoelectric properties close to the morphotropic phase boundary (MPB) between rhombohedral and tetragonal. Nevertheless, they are not environmental friendly for their lead oxide toxicity. With recent growing demand of global environmental protection, many researchers have greatly focused on lead-free ceramics to replace the lead-based ceramics [1–3].

The newly discovered lead-free $\text{Ba}(\text{Zr}_{0.2}\text{Ti}_{0.8})\text{O}_3\text{--}x(\text{Ba}_{0.7}\text{Ca}_{0.3})\text{TiO}_3$ ceramics by Liu and Ren have attracted great attention due to the excellent piezoelectric properties (with $d_{33} = 350\text{--}650$ pC/N) [4–11]. A d_{33} of 650 pC/N is obtained for the composition of $(\text{Ba}_{0.85}\text{Ca}_{0.15})(\text{Ti}_{0.9}\text{Zr}_{0.1})\text{O}_3$ (BCZT) by optimizing sintering temperature at 1540 °C [9]. However the optimal composition with high d_{33} value is

sintered at high temperature. It is well known that CeO_2 is commonly used as an additive for lead-free ceramics to further improve density and piezoelectric properties, such as BNT ceramics [12–14]. Therefore the purpose of this paper is to present the effect of CeO_2 addition on improving the sinterability and the electrical properties of BCZT ceramics. The dielectric properties of $(\text{Ba}_{0.85}\text{Ca}_{0.15})(\text{Ti}_{0.9}\text{Zr}_{0.1})\text{O}_3\text{--}x\%\text{CeO}_2$ (BCZT– $x\text{Ce}$) ceramics are measured as a function of temperature. The modification mechanism of added CeO_2 has also been discussed.

2. Experimental

Using conventional solid-state reaction technique BCZT ceramics and CeO_2 ($x = 0\text{--}0.1$ wt%) doped BCZT– $x\text{Ce}$ ceramics were prepared. Systems were prepared by using a two-step method. Raw materials of BaCO_3 (99.8%), CaCO_3 (99.0%), ZrO_2 (99.5%), and TiO_2 (99.9%) were weighed and mixed in a centrifugal mill, as well as wet-homogenized with ethanol for 8 h. The obtained powders were dried in an oven, and then calcined at 1250 °C for 2 h. Different amounts of CeO_2

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were added into the calcined powders, respectively. The resulting mixtures were further mixed with 5% poly vinyl alcohol (PVA) solution thoroughly. The obtained powders were pressed into disks specimens with a diameter of 18 mm and a thickness of 1 mm only by uniaxial pressing with a pressure of 100 Mpa. After burning off PVA, sintering was carried out at the temperature of 1350 °C for 4 h.

The crystalline structure of sintered ceramics was examined by X-ray diffraction (XRD) (Bruker D8 Advance) with Cu-K α radiation (step: 0.02°). Surface microstructure was observed by scanning electron microscopy (SEM) (JEOL, JSM-5610LV). Dielectric properties of the specimens were obtained together using an impedance analyzer (Agilent 4294A) by measuring the capacitance and loss (Cp–D) of the specimen homogenously from 20 °C to 150 °C, using with a heating rate of 60 °C h^{−1}. The planar coupling coefficient (k_p) was determined by the resonance–anti-resonance method, using Agilent 4294A impedance analyze. The piezoelectric coefficient (d_{33}) was recorded from 24-h-aged polarized samples using a quasi static piezoelectric d_{33} meter (ZJ-3AN, Institute of Acoustics, Chinese Academy of Sciences, China).

3. Results and discussion

Fig. 1 shows the X-ray diffraction (XRD) patterns of BCZT– x Ce ceramics with different contents of CeO₂ sintered at 1350 °C. As can be seen from Fig. 1(a), a single perovskite structure is observed for all sintered samples and there is no trace of second phase, which implies that Y ions diffuse into the BCZT lattices and a solid solution forms in the ceramics. Further XRD analysis is performed in the 2θ ranges from 44° to 47° as shown in Fig. 1(b). The XRD patterns exhibit the feature of peak which splits at about 44° and 47°, which indicates that coexistence of rhombohedral and tetragonal phases in the BCZT– x Ce ceramic samples with 0.0–0.1% CeO₂ at room temperature [4,9]. Meanwhile, the diffraction peaks of CeO₂ doped BCZT ceramics shift slightly to higher angles with increasing CeO₂ content. It is expected from the tolerance factors that small ions [$r(R^{3+}) < 0.087$ nm] will occupy the B site, large ions [$r(R^{3+}) > 0.094$ nm] will occupy the A site, and intermediate ions will occupy both sites with different partitioning for each ion [8]. Ce ions possibly exist in the BCZT structure in two valence states: Ce⁴⁺ in a radius of 0.094 nm, Ce³⁺ in a radius of 0.107 nm. Taking Ce ions into consideration, Ce ions should occupy the A-site of Ca²⁺ ($r = 0.1$ nm) and Ba²⁺ ($r = 0.135$ nm). According to the peaks shift to higher angles, the lattice parameter should decrease, and it is assumed that Ce ions occupy the A-site of Ba²⁺.

As shown in Fig. 2, the SEM surface micrographs of the BCZT– x Ce ceramics modified with CeO₂ were sintered at 1350 °C for 4 h. It is evident that CeO₂ addition causes a large change in the grain size. The pure BCZT ceramics have a porous microstructure with small grains, as shown in Fig. 2(a). However, when a small amount of CeO₂ is added, a dense microstructure with enlarged grains is developed [see Fig. 2(b)]. The non-uniformity of grains seems to be slightly improved with the further addition of CeO₂, as seen from

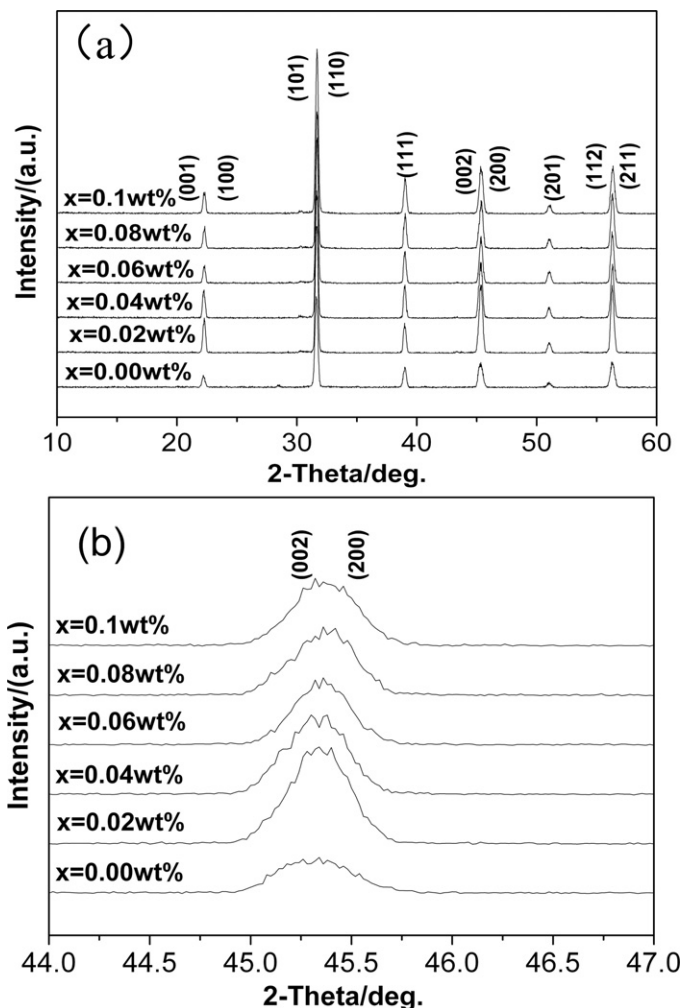


Fig. 1. X-ray diffraction patterns of BCZT– x CeO₂ ceramics sintered at 1350 °C.

Fig. 2(c)–(f). The addition of CeO₂ promotes the mobility of grain boundary to a certain degree so that a relative non-uniformity of grain can be obtained. Therefore, it is believed that a small amount of CeO₂ adding to BCZT ceramics can modify the microstructure and the grain size of the BCZT ceramics, and result in a reduction of sintering temperature.

Fig. 3 shows the piezoelectric coefficient d_{33} and planar mode electromechanical coupling coefficient k_p of the BCZT– x Ce ceramics as a function of CeO₂ content. It can be seen from Fig. 3 that with the increasing of CeO₂ content, the piezoelectric constant d_{33} and the planar coupling factor k_p increase slowly up to a maximum value 600 pC/N at 0.04% CeO₂ and 0.54 at 0.08% CeO₂, and then decrease sharply with further increase of CeO₂. It is believed that the observed high piezoelectric properties should be ascribed to the phase coexistence [4,9] and the first-order PPT occurring near room temperature should be the origin of this phase coexistence. However, with the increasing of CeO₂ content to 0.1%, Ce ions should occupy the A-site of Ba²⁺, which cause deformation in the ABO₃ lattice to make a phase transition completed at a lower temperature, resulting in a reduction of piezoelectric

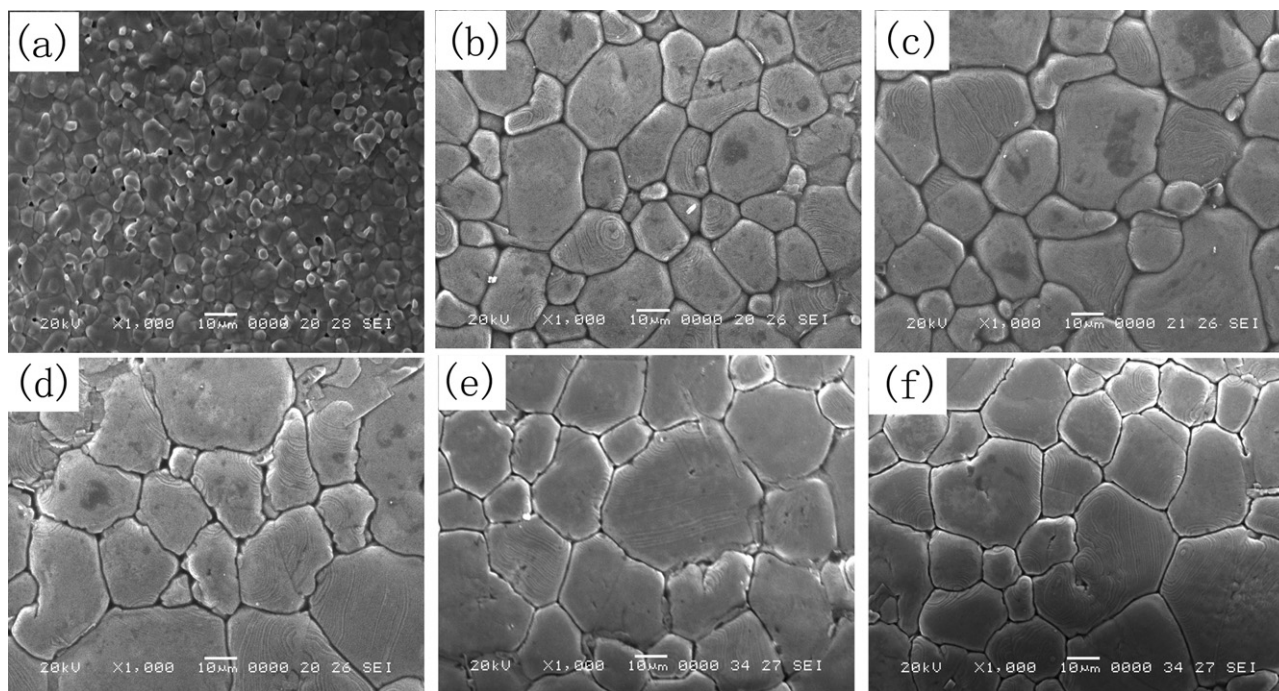


Fig. 2. SEM images of the BCZT- $x\%$ CeO₂ ceramic sintered at 1350 °C: (a) $x = 0.00$, (b) $x = 0.02$, (c) $x = 0.04$, (d) $x = 0.06$, (e) $x = 0.08$ and (f) $x = 0.1$.

coefficient d_{33} . This shows that a proper addition of CeO₂ induces a more significant piezoelectric activity, which can be mainly attributed to grain size, the densification and morphology of the ceramics.

The dielectric constant ϵ_r and dielectric loss ($\tan \delta$) for BCZT- x Ce ceramics measured at 1 kHz are shown in Fig. 4. As we can see from Fig. 4, with increasing CeO₂ content the relative dielectric constant ϵ_r (1 kHz) first decreases to a peaked value of 4464 at 0.08% CeO₂ and then descends rapidly to 3866 at 0.1% CeO₂, the maximum ϵ_r (4923) is obtained at 0.02% CeO₂. Moreover the $\tan \delta$ (at 1 kHz) distinctly decreases at first and reaches a minimum value of 0.9% at 0.08% CeO₂ addition. Generally speaking, the values of ϵ_r and $\tan \delta$ decrease with increasing of CeO₂ content.

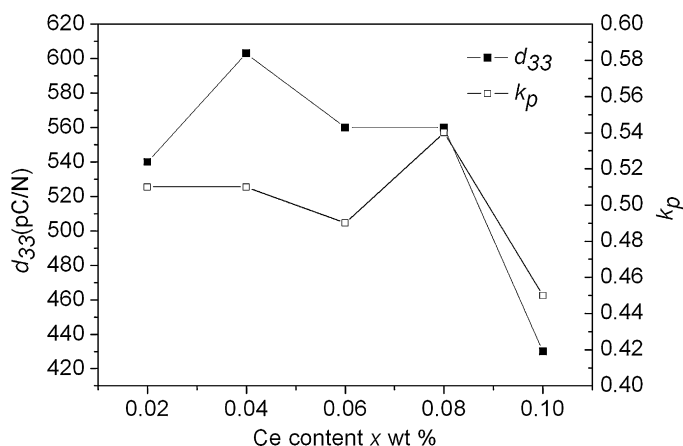


Fig. 3. The piezoelectric constant d_{33} and the planar electromechanical coefficient k_p of the BCZT- $x\%$ CeO₂ ceramics.

Fig. 5 shows the temperature dependence of the relative dielectric constant for BCZT- x Ce ceramics measured at 10 kHz. One peak is observed on the dielectric constant versus temperature curves in the measured temperature range (20–150 °C). It is considered that the peak corresponds to the polymorphic phase transitions from tetragonal to cubic phase [4,9] the transition temperature (T_c) is around 90 °C. Besides, with the increasing CeO₂ content to 0.1%, the curie temperature T_c decreases from 90 °C to 85 °C because the Ce ions [$r(\text{Ce}^{4+}) = 0.094$ nm, $r(\text{Ce}^{3+}) = 0.107$ nm] are substituted for A-site of Ba²⁺ ($r = 0.135$ nm), resulting in the change of the ceramics lattice which cause deformation in the ABO₃ lattice. When the content of CeO₂ is low, the deformation is weak and the structure of the ceramic is

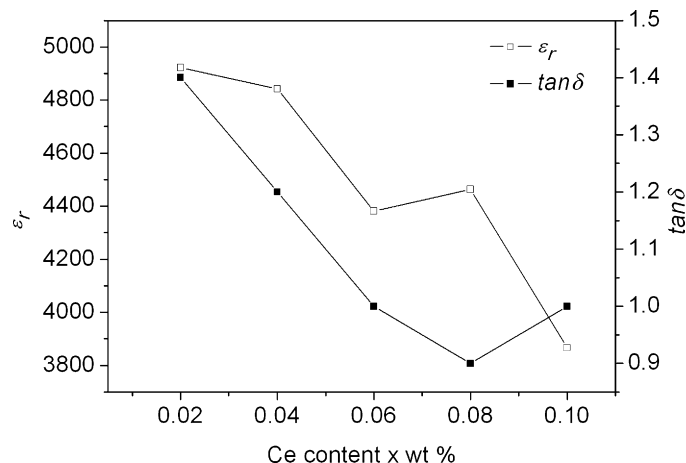


Fig. 4. The dielectric constant ϵ_r and the dielectric loss $\tan \delta$ of the BCZT- $x\%$ CeO₂ ceramics.

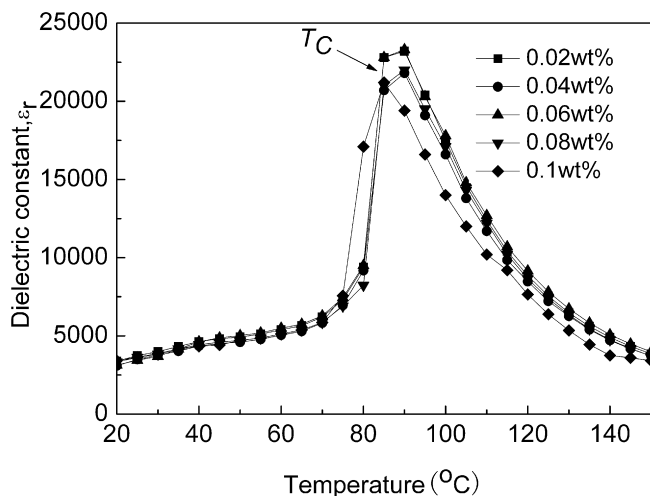


Fig. 5. Temperature dependence of dielectric constant for BCZT- $x\%$ CeO₂ ceramics measured at 10 kHz.

stable. However, when the content of CeO₂ become high, for example at 0.1%, the deformation cannot be ignored and the structure of the ceramic become non-stable. So when increasing the content of CeO₂, the structure change from tetragonal phase to cubic phase at lower temperature.

4. Conclusions

Lead-free BCZT- x Ce piezoelectric ceramics have been prepared by the solid-state reaction technique. The crystal-line phases, dielectric and piezoelectric properties have been studied in the paper. X-ray diffraction data show that CeO₂ diffuses into the lattice of BCZT with formation of a pure perovskite phase and reduction in sintering temperature. A small addition of CeO₂ can largely reduce the sintering temperature. Main piezoelectric parameters were optimized in the vicinity of $x = 0.04$ wt% with comparatively lower sintering temperature (1350 °C), showing a promising future as lead-free ceramics for low temperature applications.

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