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Large-scale fine structural alumina matrix ceramic guideway materials improved by diopside and Fe₂O₃

Changxia Liu a,b,*, Jianhua Zhang a,*, Xihua Zhang c, Junlong Sun a,b

^a Department of Mechanical Engineering, Shandong University, Jinan 250061, Shandong Province, PR China ^b Communications Institute, Ludong University, Yantai 264025, Shandong Province, PR China ^c Department of Materials Science and Engineering, Shandong University, Jinan 250061, Shandong Province, PR China Received 3 July 2006; received in revised form 25 July 2006; accepted 2 September 2006 Available online 7 November 2006

Abstract

Diopside and Fe₂O₃ were introduced in alumina matrix ceramic materials. Large-scale fine structural alumina matrix ceramic guideway materials were fabricated by the technology of pressureless sintering, during which liquid phase sintering took place and new phases such as 3Al₂O₃·2SiO₂, CaO·Al₂O₃·2SiO₂ and CaO·6Al₂O₃ were produced by the chemical reactions taking place among alumina and the additives. The hardness, the fracture toughness and the bending strength of the guideway products were tested. The influences of diopside and Fe₂O₃ additions were studied by microstructural observations and mechanical properties evaluations. Meanwhile, the expected improvement of mechanical properties compared with pure alumina was indeed observed. The fracture mechanism and porosity of large-scale fine structural alumina matrix ceramic guideway materials were analyzed.

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1. Introduction

Alumina matrix ceramic materials are widely applied in the fields of mechanical engineering [1–4], electronic technology [5], chemical engineering [6–9] and biological engineering [10]. Machine tool guideway in mechanical engineering demands for excellent wear resistance and low friction coefficient, conventional guideway materials such as cast iron (quenched include), steel, plastics, and polymeric compounds always have a short length of life. These materials may be likely to give way to fine alumina matrix ceramic materials for their virtues of high hardness, good chemical inertness, high wear resistance, low coefficient of thermal expansion and low friction coefficient. However, the brittleness of pure alumina limits its potential applications, and excellent mechanical properties are obtained always companied with a cost increase on account of the expensive second phases [11–15]. It generally

In this paper, large-scale fine structural alumina matrix ceramic guideway materials with different contents of diopside and Fe₂O₃ were fabricated by pressureless sintering technology in a furnace without any atmosphere protection. The specific objective of the study is to investigate the effects of diopside

E-mail addresses: hester5371@gmail.com (C. Liu), jhzhang@sdu.edu.cn (J. Zhang).

costs more when fabricating ceramic guideway materials owing to its large dimension. So low-cost also may be regarded as a priority when researchers design the compositions of largescale fine alumina matrix ceramic guideway materials. Diopside (MgCa(SiO₃)₂) and Fe₂O₃ just has the virtue of low-cost by contrast to other additives. The sources of α-Al₂O₃, diopside and Fe₂O₃ are in abundance in China and their price is low. Fe₂O₃ is generally used as a nucleant agent of glass-ceramics [16,17]. Diopside acts as an accessory ingredient in the sintering process and decreases the sintering temperature of alumina matrix ceramic materials. As for large-scale fine structural alumina matrix ceramic guideway products, their large dimension and complicated shape make it difficult to fabricate these ceramic materials by the process of hotpressing. So pressureless sintering technology was chosen and liquid phase sintering can take place during the process of sintering.

^{*} Corresponding authors.

and Fe₂O₃ on mechanical properties, degree of porosity and microstructures of large-scale fine structural alumina matrix ceramic guideway materials.

2. Experimental procedure

Commercial Al_2O_3 powder of high purity (99.99%) and small grain size (0.5–1 μ m) was used as the starting materials. Fe₂O₃ and diopside (MgCa(SiO₃)₂), which is composed of SiO₂ (55 wt.%), CaO (24 wt.%) and MgO (18 wt.%), were used as additives. The content of diopside and Fe₂O₃ is listed in Table 1 (the suffixes in AD_0F_0 , AD_5F_0 , AD_5F_3 , $AD_{10}F_0$ and $AD_{10}F_3$ represent the weight content of diopside and Fe₂O₃. For example, $AD_{10}F_3$ means the content of diopside and Fe₂O₃ are 10 and 3 wt.%, respectively).

First, the raw materials were blended with each other according to certain proportions and ball milled for $60 \, h$ in an alcohol medium to obtain an homogeneous mixture. Second, the slurry was dried in vacuum and screened. Third, the green bodies of guideway ($1800 \, \text{mm} \times 150 \, \text{mm} \times 50 \, \text{mm}$) were shaped by slip casting process and then dense green bodies of guideway are obtained by using isostatic cool pressing technology in rubber molds. Lastly, pressureless sintering was used to sinter these dense green bodies in a furnace in air. The green bodies of guideway were heated up to $1500 \, ^{\circ}\text{C}$ according to a temperature gradient [18], as listed in Table 2.

The sintered guideway products were cut into specimens by an inside diameter slicer. Standard test pieces $(3 \text{ mm} \times 4 \text{ mm} \times 36 \text{ mm})$ were obtained through rough grinding, finish grinding with diamond wheels and polishing. Three-point bending mode was used to measure the bending strength on an electronic universal experimental instrument (WD-10) with a span of 20 mm at a crosshead speed of 0.5 mm/min. Twelve specimens, chosen from one guideway product with the same compositions, were used for measuring the bending strength in air at room temperature. The bending strength was calculated by the following formula [19]:

$$\sigma_{\rm f} = \frac{3PL}{2hh^2} \tag{1}$$

where σ_f is the bending strength (MPa), P is the load (N) under which the samples break, b and h are width and height (mm), respectively, and L is the span (mm).

Vickers hardness was measured on polished surface with a load of 9.8 N for 5 s with a micro-hardness tester (MH-6). Fracture toughness measurement was performed using indentation method with a hardness tester (Hv-120), and results were obtained by the formula proposed by Cook and Lawn [20].

Compositions of alumina matrix ceramic guideway materials

Specimens	Compositions (wt.%)
AD_0F_0	100% Al ₂ O ₃
AD_5F_0	5% diopside + $95%$ Al ₂ O ₃
AD_5F_3	5% diopside + 92% $Al_2O_3 + 3\% Fe_2O_3$
$AD_{10}F_0$	10% diopside + $90%$ Al ₂ O ₃
$AD_{10}F_3$	10% diopside + 87% Al_2O_3 + 3% Fe_2O_3

Table 2
Temperature gradient of pressureless sintering process (115 h) [18]

Temperature (°C)	Speed of heating up and cooling (°C/h)		
20–300	10		
300-600	20		
600–900	30		
900-1500	40		
1500	3 h holding		
1500-1000	20		
1000-50	50		
50	Open the furnace door		

XRD (D/max-2400) analysis was undertaken to identify the crystal phases after sintering. The microstructures of the specimens were studied on fracture surfaces and polished surfaces by scanning electron microscopy (HITACHI S-570).

3. Results and discussion

3.1. X-ray diffraction phase analysis

The X-ray diffraction phase analysis of AD_5F_0 and AD_5F_3 specimens are shown in Figs. 1 and 2, respectively. It is clear from Figs. 1 and 2 that there exist in AD_5F_0 and AD_5F_3 the newly formed $3Al_2O_3 \cdot 2SiO_2$, $CaO \cdot Al_2O_3 \cdot 2SiO_2$ and $CaO \cdot 6Al_2O_3$ phases. Theoretically there should exist $MgO \cdot Al_2O_3$ in the composites, but XRD analysis did not show the occurrence of

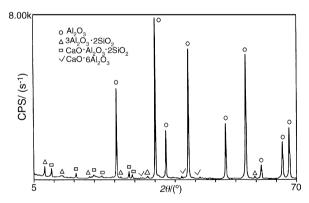


Fig. 1. X-ray diffraction phase analysis of AD₅F₀ specimen.

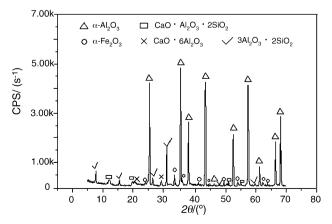


Fig. 2. X-ray diffraction phase analysis of AD₅F₃ specimen.

Table 3
Mechanical properties of alumina matrix ceramic guideway materials

Specimens	Bending strength (MPa)	Fracture toughness (MPa m ^{1/2})	Hardness (GPa)	Relative density (%)
AD_0F_0	113.0	3.27	5.6	90.2
AD_5F_0	320.6	4.62	12.0	92.7
AD_5F_3	335.5	4.51	12.8	92.9
$AD_{10}F_0$	341.0	4.10	13.5	94.0
$AD_{10}F_3$	349.3	4.32	14.4	94.3

MgO·Al $_2$ O $_3$. This is because the content of diopside is only 5 wt.% in the composites, and that of MgO is 18 wt.% in diopside, so the percent of MgO in the composites is only 0.9 wt.% and thus the amount of MgO·Al $_2$ O $_3$, produced by the reaction taking place between MgO and Al $_2$ O $_3$, is too small to be detected by XRD analysis.

The sintering temperature (1500 $^{\circ}$ C) is higher than the melting point of diopside (1300–1390 $^{\circ}$ C). Hence liquid phase sintering takes place and the following chemical reactions, yielding mullite, anorthite and CaO-6Al₂O₃, may occur during the sintering process:

$$3Al_2O_3 + 2SiO_2 \rightarrow 3Al_2O_3 \cdot 2SiO_2 \tag{2}$$

$$CaO + Al_2O_3 + 2SiO_2 \rightarrow CaO \cdot Al_2O_3 \cdot 2SiO_2$$
 (3)

$$6Al_2O_3 + CaO \rightarrow CaO \cdot 6Al_2O_3 \tag{4}$$

Based on thermodynamic analysis of the reactions, we can get the following rankings [21]:

$$\Delta G_{T2}^{\theta} < 0$$
, $\Delta G_{T3}^{\theta} < 0$, $\Delta G_{T4}^{\theta} < 0$

where ΔG_{T2}^{θ} , ΔG_{T3}^{θ} and ΔG_{T4}^{θ} represent the Gibbs free energy of the Eqs. (2)–(4), respectively. XRD analysis shows that mullite, anorthite and CaO·6Al₂O₃ actually exist in specimen of AD₅F₀ and AD₅F₃ as previously mentioned. The existing of mullite in large-scale fine structural alumina matrix ceramic guideway materials may be helpful to make the composite has a high potential for wear resistance applications [1].

3.2. Mechanical properties

Mechanical properties of pressureless sintered large-scale fine structural alumina matrix ceramic guideway materials are listed in Table 3. As seen from Table 3, the bending strength, fracture toughness and hardness of pressureless sintered pure alumina are 113.0 MPa, 3.27 MPa m^{1/2} and 5.6 GPa, respectively. Addition of diopside and Fe₂O₃ significantly improves the performances of pressureless sintered guideway composites. The bending strength and hardness of guideway composites increase with increasing the amount of diopside, and addition of Fe₂O₃ promotes this trend of improvement when the content of diopside is identical, then reach their maximum value of 349.3 MPa and 14.4 MPa m^{1/2}, respectively. This is because diopside and Fe₂O₃ all improve the densification rate of the composites (relative density in Table 3). Kim et al. [22] has observed that the mechanical properties of the composites increased with the increasing of the relative density of sintered specimens. So addition of diopside and Fe₂O₃ may be helpful to improve the mechanical properties of large-scale fine structural alumina matrix ceramic guideway materials. The variation of fracture toughness of guideway composites changing with content of diopside and Fe₂O₃ is a little indistinct, it reaches its maximum value of 4.62 MPa m^{1/2} in AD₅F₀ specimen. The mechanical properties of guideway composites are also to a great extent relevant to their microstructures, which will be discussed in the following section.

It is obvious that the fabricated large-scale fine structural alumina matrix ceramic guideway materials, pressureless sintered at 1500 °C for 3 h in air, exhibit significant improvements in mechanical properties compared to pure Al_2O_3 . Composite with the addition of 10 wt.% diopside and 3 wt.% Fe_2O_3 ($AD_{10}F_3$ specimen) shows better comprehensive performances, the bending strength, fracture toughness and the hardness of the composite reach 349.3 MPa, 4.32 MPa m^{1/2} and 14.4 GPa, respectively, which are enhanced by 209.1%, 32.1% and 157.1%, respectively, with respect to pure Al_2O_3 pressureless sintered under the same conditions.

3.3. Analysis of microstructures

SEM photomicrograph on fracture surface of pure alumina pressureless sintered at $1500 \,^{\circ}\text{C}$ is shown in Fig. 3, those of AD_5F_0 , $AD_{10}F_0$ and $AD_{10}F_3$ are shown in Figs. 4–6,

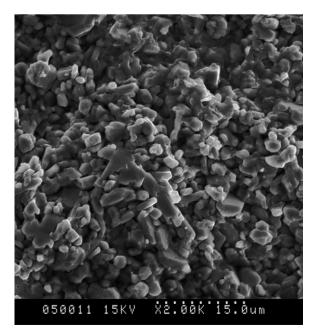


Fig. 3. SEM photomicrograph on fracture surface of pure Al₂O₃.

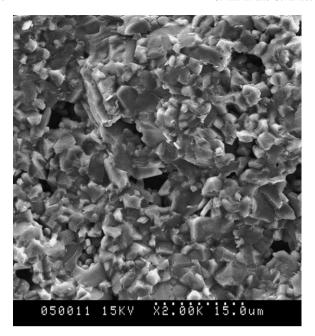


Fig. 4. SEM photomicrograph on fracture surface of AD₅F₀ specimen.

respectively. As seen from Fig. 3, the grain shapes of pure alumina are irregular and there appears abnormal growth. While in Figs. 4–6, the guideway composites, sintered under the same conditions, show a homogeneous distribution of alumina grains and additive particles. The microstructures of alumina matrix ceramic guideway composites are finer than that of pure alumina.

The fracture mode of pure alumina is mainly intergranular failure. While that of AD_5F_0 , $AD_{10}F_0$ and $AD_{10}F_3$ show a combination of intergranular failure and transgranular failure. Further addition of diopside makes the occurrence of transgranular failure more than that of intergranular failure

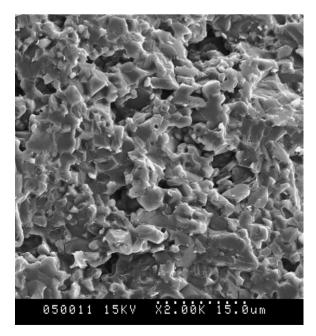


Fig. 5. SEM photomicrograph on fracture surface of $AD_{10}F_0$ specimen.

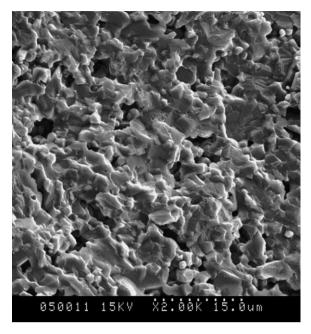


Fig. 6. SEM photomicrograph on fracture surface of AD₁₀F₃ specimen.

(Figs. 4 and 5). The microstructure of $AD_{10}F_3$ is analogous to that of $AD_{10}F_0$ except that $AD_{10}F_3$ specimen shows a little more homogeneous and finer microstructure than $AD_{10}F_0$ (Figs. 5 and 6). Obviously, when the amount of diopside is identical, addition of Fe_2O_3 makes the microstructure of guideway composites a little more homogeneous and finer, which may bring on the improvements in bending strength of large-scale fine structural alumina matrix ceramic guideway materials.

As seen from the microstructures of pure alumina, AD_5F_0 , $AD_{10}F_0$ and $AD_{10}F_3$, the bondings of grains become stronger when the fracture mode turns to the combination of transgranular failure and intergranular failure. In other words,

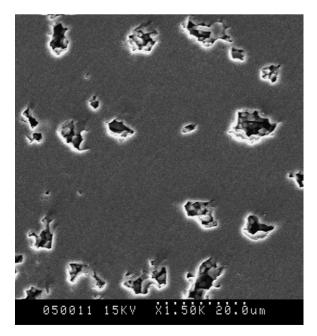


Fig. 7. SEM photomicrograph on polished surface of AD₅F₀ specimen.

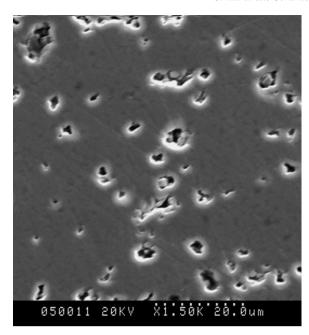


Fig. 8. SEM photomicrograph on polished surface of AD₅F₃ specimen.

a fracture mode changing from intergranular failure to the combination of transgranular failure and intergranular failure is observed and may bring on a significant increase in bending strength and fracture toughness. The main cause is that the grain boundaries of large-scale fine structural alumina matrix ceramic guideway materials are strengthened owing to the addition of diopside and Fe_2O_3 .

Figs. 7–9 respectively show the SEM photomicrographs on polished surfaces of AD_5F_0 , AD_5F_3 and $AD_{10}F_0$ specimens. As seen from these figures, there exist apparent pores on the polished surface of large-scale fine structural alumina matrix ceramic guideway materials, which may be difficult to avoid in

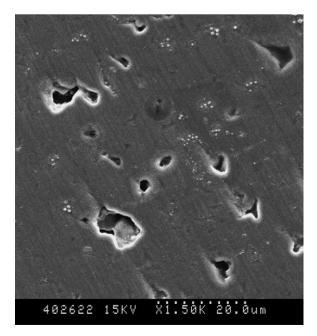


Fig. 9. SEM photomicrograph on polished surface of AD₁₀F₀ specimen.

pressureless sintering technology. Pores on polished surface of $AD_{10}F_0$ specimen (Fig. 9) are much fewer than that in AD_5F_0 and AD_5F_3 specimens (Figs. 7 and 8), which makes it clear that the addition of diopside decreases the amount of pores in large-scale fine structural alumina matrix ceramic guideway materials. Smaller gauge of pores are found in AD_5F_3 specimen (Fig. 8) by contrast to that in AD_5F_0 specimen (Fig. 7), indicating that the addition of Fe_2O_3 disperses pores in the guideway composites, incorporating into the same content of diopside.

4. Conclusion

Large-scale fine structural alumina matrix ceramic guideway materials were fabricated by pressureless sintering technology according to a temperature gradient. Liquid phase sintering takes place during the sintering process, during which new phases such as mullite, anorthite and $\text{CaO} \cdot 6\text{Al}_2\text{O}_3$ were produced by the chemical reactions taking place among alumina and diopside. Addition of diopside and Fe_2O_3 significantly improves the performances of pressureless sintered guideway composites. The bending strength, fracture toughness and the hardness of AD_{10}F_3 specimen are 349.3 MPa, 4.32 MPa m^{1/2} and 14.4 GPa, respectively, which are enhanced by 209.1%, 32.1% and 157.1%, respectively, with respect to pure Al_2O_3 pressureless sintered under the same conditions.

The shapes of pure alumina grains, unevenly distributed, are irregular and abnormal growth appears. Composites with the addition of diopside and Fe₂O₃ sintered under the same conditions show more homogeneous and finer microstructures. When the amount of diopside is identical, addition of Fe₂O₃ makes the microstructures of guideway composites a little more homogeneous and finer. The fracture mode of pure Al₂O₃ is mainly intergranular failure. While that of composites with addition of diopside and Fe₂O₃ is the combination of transgranular failure and intergranular failure, which may bring on the increases in bending strength and fracture toughness. Addition of diopside decreases the amount of pores in large-scale fine structural alumina matrix ceramic guideway materials. A small amount of Fe₂O₃ addition makes the pores smaller and more evenly distributed in the guideway composites, which may also contribute to the improvements in mechanical properties of large-scale fine structural alumina matrix ceramic guideway materials.

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References

[1] E. Medvedovski, Alumina–mullite ceramics for structural applications, Ceram. Int. 32 (4) (2006) 369–375.

- [2] D.W. Lee, B.K. Kim, Nanostructured Cu–Al₂O₃ composite produced by thermochemical process for electrode application, Mater. Lett. 58 (3/4) (2004) 378–383.
- [3] G.S. Huang, Y. Xie, X.L. Wu, et al., Formation mechanism of individual alumina nanotubes wrapping metal (Cu and Fe) nanowires, J. Cryst. Growth 289 (1) (2006) 295–298.
- [4] M. Jung, H.-G. Kim, J.-K. Lee, et al., EDLC characteristics of CNTs grown on nanoporous alumina templates, Electrochim. Acta 50 (2/3) (2004) 857–862.
- [5] V. Viswabaskaran, F.D. Gnanam, M. Balasubramanian, Mullite from clayreactive alumina for insulating substrate application, Appl. Clay Sci. 25 (1/2) (2004) 29–35.
- [6] A. Aguilera, V. Jayaraman, S. Sanagapalli, et al., Porous alumina templates and nanostructured CdS for thin film solar cell applications, Solar Energy Mater. Solar Cells 90 (6) (2006) 713–726.
- [7] S. Alini, A. Bottino, G. Capannelli, et al., Preparation and characterisation of Rh/Al₂O₃ catalysts and their application in the adiponitrile partial hydrogenation and styrene hydroformylation, Appl. Catal. A: Gen. 292 (2005) 105–112
- [8] Y. Zhang, M. Koike, R. Yang, et al., Multi-functional alumina-silica bimodal pore catalyst and its application for Fischer-Tropsch synthesis, Appl. Catal. A: Gen. 292 (2005) 252–258.
- [9] A. Garron, K. Lázár, F. Epron, Effect of the support on tin distribution in Pd-Sn/Al₂O₃ and Pd-Sn/SiO₂ catalysts for application in water denitration, Appl. Catal. B: Environ. 59 (1/2) (2005) 57–69.
- [10] B. Kasprzyk-Hordern, U. Raczyk-Stanisławiak, J. Świetlik, et al., Catalytic ozonation of natural organic matter on alumina, Appl. Catal. B: Environ. 62 (3/4) (2006) 345–358.
- [11] S.I. Cha, K.T. Kim, K.H. Lee, et al., Strengthening and toughening of carbon nanotube reinforced alumina nanocomposite fabricated by molecular level mixing process, Scripta Mater. 53 (7) (2005) 793–797.

- [12] S.K.C. Pillai, B. Baron, M.J. Pomeroy, et al., Effect of oxide dopants on densification, microstructure and mechanical properties of alumina–silicon carbide nanocomposite ceramics prepared by pressureless sintering, J. Eur. Ceram. Soc. 24 (2004) 3317–3326.
- [13] W. Nakao, M. Ono, S.-K. Lee, et al., Critical crack-healing condition for SiC whisker reinforced alumina under stress, J. Eur. Ceram. Soc. 25 (16) (2005) 3649–3655.
- [14] P.C. Ostertag, Influence of fiber and grain bridging on crack profiles in SiC fiber-reinforced alumina-matrix composites, Mater. Sci. Eng. A 260 (1/2) (1999) 124–131.
- [15] E. Laarz, M. Carlsson, B. Vivien, et al., Colloidal processing of Al₂O₃-based composites reinforced with TiN and TiC particulates, whiskers and nanoparticles, J. Eur. Ceram. Soc. 21 (8) (2001) 1027–1035.
- [16] P. Alizadeh, B. Eftekhary Yekta, A. Gervei, Effect of Fe₂O₃ addition on the sinterability and machinability of glass-ceramics in the system MgO– CaO–SiO₂–P₂O₅, J. Eur. Ceram. Soc. 24 (2004) 3529–3533.
- [17] M. Rezvani, B. Eftekhari-Yekta, M. Solati-Hashjin, et al., Effect of Cr₂O₃, Fe₂O₃ and TiO₂ nucleants on the crystallization behaviour of SiO₂– Al₂O₃–CaO–MgO (R₂O) glass-ceramics, Ceram. Int. 31 (2005) 75–80.
- [18] X.H. Zhang, C.X. Liu, J.H. Zhang, Research on the sintering process and performance of large-scale alumina matrix ceramic products, Mater. Sci. Forum 471/472 (2004) 821–824.
- [19] Z.H. Jin, J.Q. Gao, G.H. Qiao, Engineering Ceramics [M], Publishing House of Xian Jiao Tong University, Xian, 2000 (in Chinese).
- [20] R.F. Cook, B.R. Lawn, A modified indentation toughness technique, J. Am. Ceram. Soc. 66 (11) (1983) 200–201.
- [21] D.L. Ye, J.H. Hu, Handbook of The Thermodynamic Data of Inorganic Substances, Metallurgical Industry Press, Peking, China, 2002pp. 57–1061.
- [22] H.W. Kim, Y.H. Koh, H.E. Kim, Densification and mechanical properties of B₄C with Al₂O₃ as a sintering aid, J. Am. Ceram. Soc. 83 (11) (2000) 2863–2865.